

Global Analysis of the NH_2 Radical Rotational, Vibrational, and Electronic Transitions

M.-A. Martin-Drumel, O. Pirali, and L. H. Coudert

ISMO, Universités Paris-Sud & Paris-Saclay, Orsay, France

ISMS, June 17–21, 2019



Introduction

- The NH_2 radical, first observed a long time ago,¹ is a benchmark molecule for the [Renner-Teller effect](#)²

¹Herzberg & Ramsay, *J. Chem. Phys.* **20** (1952) 347

²Dressler and Ramsay, *Phil. Trans. R. Soc. A* **25** (1959) 553

³van Dishoeck, Jansen, Schilke, & Phillips, *Astr. Phys. J.* **416** (1993) L83

⁴Kawakita, Ayani, & Kawabata, *Publ. Astr. Soc. Japan* **52** (2000) 925

⁵Jungen & Merer, *Molec. Phys.* **40** (1980) 1; Duxbury & Dixon, *Molec. Phys.* **43** (1981) 255; and Jensen, Odaka, Kraemer, Hirano, & Bunker, *Spectrochim. Acta Part A* **58** (2002) 763

⁶Jensen, Kraemer, & Bunker, *Molec. Phys.* **101** (2003) 613

Introduction

- The NH_2 radical, first observed a long time ago,¹ is a benchmark molecule for the **Renner-Teller effect**²
- Molecule of **astrophysical interest** detected in the ISM and in comets^{3,4}

¹Herzberg & Ramsay, *J. Chem. Phys.* **20** (1952) 347

²Dressler and Ramsay, *Phil. Trans. R. Soc. A* **25** (1959) 553

³van Dishoeck, Jansen, Schilke, & Phillips, *Astr. Phys. J.* **416** (1993) L83

⁴Kawakita, Ayani, & Kawabata, *Publ. Astr. Soc. Japan* **52** (2000) 925

⁵Jungen & Merer, *Molec. Phys.* **40** (1980) 1; Duxbury & Dixon, *Molec. Phys.* **43** (1981) 255; and Jensen, Odaka, Kraemer, Hirano, & Bunker, *Spectrochim. Acta Part A* **58** (2002) 763

⁶Jensen, Kraemer, & Bunker, *Molec. Phys.* **101** (2003) 613

Introduction

- The NH_2 radical, first observed a long time ago,¹ is a benchmark molecule for the [Renner-Teller effect](#)²
- Molecule of [astrophysical interest](#) detected in the ISM and in comets^{3,4}
- Large number of [experimental](#) and [theoretical](#)⁵ investigations

¹Herzberg & Ramsay, *J. Chem. Phys.* **20** (1952) 347

²Dressler and Ramsay, *Phil. Trans. R. Soc. A* **25** (1959) 553

³van Dishoeck, Jansen, Schilke, & Phillips, *Astr. Phys. J.* **416** (1993) L83

⁴Kawakita, Ayani, & Kawabata, *Publ. Astr. Soc. Japan* **52** (2000) 925

⁵Jungen & Merer, *Molec. Phys.* **40** (1980) 1; Duxbury & Dixon, *Molec. Phys.* **43** (1981) 255; and Jensen, Odaka, Kraemer, Hirano, & Bunker, *Spectrochim. Acta Part A* **58** (2002) 763

⁶Jensen, Kraemer, & Bunker, *Molec. Phys.* **101** (2003) 613

Introduction

- The NH_2 radical, first observed a long time ago,¹ is a benchmark molecule for the **Renner-Teller effect**²
- Molecule of **astrophysical interest** detected in the ISM and in comets^{3,4}
- Large number of **experimental** and **theoretical**⁵ investigations
- Only one global analysis carried out⁶

¹Herzberg & Ramsay, *J. Chem. Phys.* **20** (1952) 347

²Dressler and Ramsay, *Phil. Trans. R. Soc. A* **25** (1959) 553

³van Dishoeck, Jansen, Schilke, & Phillips, *Astr. Phys. J.* **416** (1993) L83

⁴Kawakita, Ayani, & Kawabata, *Publ. Astr. Soc. Japan* **52** (2000) 925

⁵Jungen & Merer, *Molec. Phys.* **40** (1980) 1; Duxbury & Dixon, *Molec. Phys.* **43** (1981) 255; and Jensen, Odaka, Kraemer, Hirano, & Bunker, *Spectrochim. Acta Part A* **58** (2002) 763

⁶Jensen, Kraemer, & Bunker, *Molec. Phys.* **101** (2003) 613

Introduction-*continued*

Two analyses of the NH_2 data are carried out

Introduction-*continued*

Two analyses of the NH_2 data are carried out

Analysis	I	II
Data set	$J \leq 4$ levels up to $14\,000\text{ cm}^{-1}$	$J \leq 25$ lines Ground & (010) states
Approach	Exact 3-D + Rot.	Effective 1-D + Rot.
Effect	Renner-Teller	Anomalous distortion

Outline

- The available data

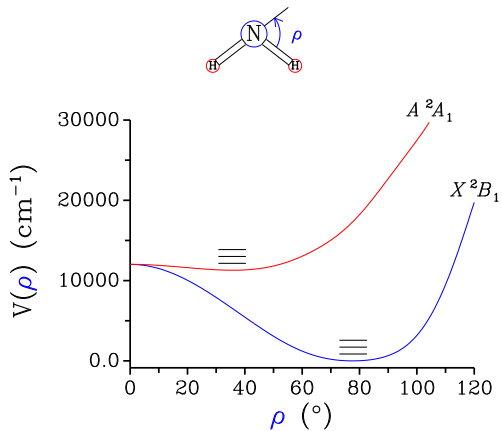
Outline

- The available data
- Analysis I
 - ▶ Theoretical approach
 - ▶ Results

Outline

- The available data
- Analysis I
 - ▶ Theoretical approach
 - ▶ Results
- Analysis II
 - ▶ Theoretical approach
 - ▶ Results

Simplified energy level diagram



Available data

X^2B_1		A^2A_1	
(000)	0	(000)	11 122
(010)	1 497	(010)	12 280
(020)	2 961	(020)	13 619
(100)	3 219		
	⋮		
(0,10,0)	13 448		
	⋮		
		(120)	17 073
<hr/>		<hr/>	
21 states		6 states	

¹Burkholder, Howard, & McKellar, *J. Molec. Spectrosc.* **127** (1988) 415; Morino & Kawaguchi, *ibid.* **182** (1997) 428; and Martin-Drumel, Pirali, and Vervloet, *J. Phys. Chem. A* **118** (2014) 1331

²Ross, Birss, Vervloet, & Ramsay, *J. Molec. Spectrosc.* **129** (1988) 436; and Hadj Bachir, Huet, Destombes, and Vervloet, *ibid.* **193** (1999) 326

³McKellar, Vervloet, Burkholder, and Howard, *J. Molec. Spectrosc.* **142** (1990) 319

Available data

X^2B_1		A^2A_1	
(000)	0 ¹	(000)	11 122 ²
(010)	1 497 ¹	(010)	12 280 ²
(020)	2 961 ³	(020)	13 619 ²
(100)	3 219 ³		
	⋮		
(0,10,0)	13 448 ²	(120)	17 073 ²
<hr/>		<hr/>	
21 states		6 states	

¹Burkholder, Howard, & McKellar, *J. Molec. Spectrosc.* **127** (1988) 415; Morino & Kawaguchi, *ibid.* **182** (1997) 428; and Martin-Drumel, Pirali, and Vervloet, *J. Phys. Chem. A* **118** (2014) 1331

²Ross, Birss, Vervloet, & Ramsay, *J. Molec. Spectrosc.* **129** (1988) 436; and Hadj Bachir, Huet, Destombes, and Vervloet, *ibid.* **193** (1999) 326

³McKellar, Vervloet, Burkholder, and Howard, *J. Molec. Spectrosc.* **142** (1990) 319

Rovibronic basis set functions

The rovibronic wavefunctions are expanded using^{1,2}

$$\chi^+(R_1, R_2, t) | +\Lambda \rangle \quad \& \quad \chi^-(R_1, R_2, t) | -\Lambda \rangle$$

¹Mitrushchenkov, *J. Chem. Phys.* **136** (2012) 024108

²Coudert, Gans, Holzmeier, Loison, Garcia, Alcaraz, Lopes, & Röder, *J. Chem. Phys.* **149** (2018) 224304

Rovibronic basis set functions

The rovibronic wavefunctions are expanded using^{1,2}

$$\chi^+(R_1, R_2, t) |+\Lambda\rangle \quad \& \quad \chi^-(R_1, R_2, t) |-\Lambda\rangle$$

where

$$\left\{ \begin{array}{ll} \chi^+, \chi^- & \text{vibrational wavefunctions} \\ |+\Lambda\rangle, |-\Lambda\rangle & \text{eigenfunctions of } L_z \\ R_1, R_2, t = \cos \tilde{\theta} & \text{Radau coordinates} \end{array} \right.$$

¹Mitrushchenkov, *J. Chem. Phys.* **136** (2012) 024108

²Coudert, Gans, Holzmeier, Loison, Garcia, Alcaraz, Lopes, & Röder, *J. Chem. Phys.* **149** (2018) 224304

Taking the singularities into account

The vibrational wavefunctions are expanded¹

$$\chi^{\pm}(R_1, R_2, t) = (1 - t)^{\frac{\alpha^{\pm}}{2}} (1 + t)^{\frac{\beta^{\pm}}{2}} \sum_{n=0} P_n^{(\alpha^{\pm}, \beta^{\pm})}(t) \phi_n(R_1, R_2)$$

¹Gutlé & Coudert, *J. Mol. Spectrosc.* **273** (2012) 44

²Coudert, *J. Mol. Spectrosc.* **154** (1992) 427; and Coudert, *ibid.* **165** (1994) 406

Taking the singularities into account

The vibrational wavefunctions are expanded¹

$$\chi^{\pm}(R_1, R_2, t) = (1 - t)^{\frac{\alpha^{\pm}}{2}} (1 + t)^{\frac{\beta^{\pm}}{2}} \sum_{n=0} P_n^{(\alpha^{\pm}, \beta^{\pm})}(t) \phi_n(R_1, R_2)$$

Singularity at the **linear** configuration and the **Renner** coupling are taken into account

¹Gutlé & Coudert, *J. Mol. Spectrosc.* **273** (2012) 44

²Coudert, *J. Mol. Spectrosc.* **154** (1992) 427; and Coudert, *ibid.* **165** (1994) 406

Taking the singularities into account

The vibrational wavefunctions are expanded¹

$$\chi^{\pm}(R_1, R_2, t) = (1 - t)^{\frac{\alpha^{\pm}}{2}} (1 + t)^{\frac{\beta^{\pm}}{2}} \sum_{n=0} P_n^{(\alpha^{\pm}, \beta^{\pm})}(t) \phi_n(R_1, R_2)$$

Singularity at the **linear** configuration and the **Renner** coupling are taken into account

Vibrational matrix elements are evaluated using **Gaussian quadratures**

¹Gutlé & Coudert, *J. Mol. Spectrosc.* **273** (2012) 44

²Coudert, *J. Mol. Spectrosc.* **154** (1992) 427; and Coudert, *ibid.* **165** (1994) 406

Potential energy surface

For the $X = \begin{cases} B_1 & \text{lower} \\ A_1 & \text{upper} \end{cases}$ electronic substates¹

$$V^X(r_1, r_3, \rho) = V_0^X(\rho) + \sum_j F_j^X(\rho) y_j + \sum_{j \leq k} F_{jk}^X(\rho) y_j y_k + \dots$$

where $j, k = 1$ or 3 , $y_j = 1 - \exp[-a(r_j - r_0)]$, and

¹Jensen, Odaka, Kraemer, Hirano, & Bunker, *Spectrochim. Acta Part A* **58** (2002) 763; and Jensen, Kraemer, & Bunker, *Molec. Phys.* **101** (2003) 613

Potential energy surface

For the $X = \begin{cases} B_1 & \text{lower} \\ A_1 & \text{upper} \end{cases}$ electronic substates¹

$$V^X(r_1, r_3, \rho) = V_0^X(\rho) + \sum_j F_j^X(\rho) y_j + \sum_{j \leq k} F_{jk}^X(\rho) y_j y_k + \dots$$

where $j, k = 1$ or 3 , $y_j = 1 - \exp[-a(r_j - r_0)]$, and

$$F_{jk}^X(\rho) = f_{jk} + \sum_{i=1}^8 f_{jk}^{i,X} (1 - \cos \rho)^i$$

¹Jensen, Odaka, Kraemer, Hirano, & Bunker, *Spectrochim. Acta Part A* **58** (2002) 763; and Jensen, Kraemer, & Bunker, *Molec. Phys.* **101** (2003) 613

Potential energy surface

For the $X = \begin{cases} B_1 & \text{lower} \\ A_1 & \text{upper} \end{cases}$ electronic substates¹

$$V^X(r_1, r_3, \rho) = V_0^X(\rho) + \sum_j F_j^X(\rho) y_j + \sum_{j \leq k} F_{jk}^X(\rho) y_j y_k + \dots$$

where $j, k = 1$ or 3 , $y_j = 1 - \exp[-a(r_j - r_0)]$, and

$$F_{jk}^X(\rho) = f_{jk} + \sum_{i=1}^8 f_{jk}^{i,X} (1 - \cos \rho)^i$$

f_{jk} and $f_{jk}^{i,X}$ are adjusted to the data.

¹Jensen, Odaka, Kraemer, Hirano, & Bunker, *Spectrochim. Acta Part A* **58** (2002) 763; and Jensen, Kraemer, & Bunker, *Molec. Phys.* **101** (2003) 613

Data set & analysis results

Experimental energy levels up to $J = 4$ and $\left\{ \begin{array}{l} (310) \tilde{X}^2B_1 \\ (010) \tilde{A}^2A_1 \end{array} \right.$

¹McKellar, Vervloet, Burkholder, and Howard, *J. Molec. Spectrosc.* **142** (1990) 319

²Vervloet, *Molec. Phys.* **63** (1988) 433

³Hadj Bachir, Huet, Destombes, and Vervloet, *J. Molec. Spectrosc.* **193** (1999) 326

Data set & analysis results

Experimental energy levels up to $J = 4$ and $\left\{ \begin{array}{l} (310) \tilde{X}^2B_1 \\ (010) \tilde{A}^2A_1 \end{array} \right.$

24 vibrational states & 753 rotational energy levels¹⁻³

¹McKellar, Vervloet, Burkholder, and Howard, *J. Molec. Spectrosc.* **142** (1990) 319

²Vervloet, *Molec. Phys.* **63** (1988) 433

³Hadj Bachir, Huet, Destombes, and Vervloet, *J. Molec. Spectrosc.* **193** (1999) 326

Data set & analysis results

Experimental energy levels up to $J = 4$ and $\left\{ \begin{array}{l} (310) \tilde{X}^2B_1 \\ (010) \tilde{A}^2A_1 \end{array} \right.$

24 vibrational states & 753 rotational energy levels¹⁻³

Varying 20 potential energy parameters, level energies were reproduced with an RMS of 2 cm⁻¹

¹McKellar, Vervloet, Burkholder, and Howard, *J. Molec. Spectrosc.* **142** (1990) 319

²Vervloet, *Molec. Phys.* **63** (1988) 433

³Hadj Bachir, Huet, Destombes, and Vervloet, *J. Molec. Spectrosc.* **193** (1999) 326

Observed minus calculated table (000) of \tilde{X}^2B_1

J	K_a	K_c	$F_{1,2}$	Obs.	O-C	J	K_a	K_c	$F_{1,2}$	Obs.	O-C
1	0	1	F_1	21.109	0.140	3	2	1	F_1	184.612	0.697
1	0	1	F_2	21.143	0.174	3	2	1	F_2	185.023	0.497
1	1	1	F_1	31.773	0.044	3	3	1	F_1	243.849	-0.215
1	1	1	F_2	32.005	-0.061	3	3	1	F_2	244.650	-0.396
1	1	0	F_1	36.535	0.141	3	3	0	F_1	244.064	-0.206
1	1	0	F_2	36.800	0.069	3	3	0	F_2	244.872	-0.378
2	0	2	F_1	62.042	0.370	4	0	4	F_1	195.685	0.862
2	0	2	F_2	62.099	0.412	4	0	4	F_2	195.780	0.838
2	1	2	F_1	69.263	0.251	4	1	4	F_1	197.560	0.805
2	1	2	F_2	69.411	0.062	4	1	4	F_2	197.674	0.570
2	1	1	F_1	83.536	0.550	4	1	3	F_1	243.460	1.711
2	1	1	F_2	83.740	0.417	4	1	3	F_2	243.672	1.555
2	2	1	F_1	115.398	-0.003	4	2	3	F_1	262.165	0.970
2	2	1	F_2	115.927	-0.141	4	2	3	F_2	262.523	0.655
2	2	0	F_1	116.669	0.056	4	2	2	F_1	277.231	1.558
2	2	0	F_2	117.196	-0.069	4	2	2	F_2	277.594	1.366
3	0	3	F_1	120.742	0.615	4	3	2	F_1	330.403	0.390
3	0	3	F_2	120.822	0.638	4	3	2	F_2	331.070	0.081
3	1	3	F_1	124.727	0.509	4	3	1	F_1	331.875	0.500
3	1	3	F_2	124.849	0.291	4	3	1	F_2	332.537	0.203
3	1	2	F_1	153.001	1.099	4	4	1	F_1	417.347	-0.612
3	1	2	F_2	153.204	0.957	4	4	1	F_2	418.427	-0.816
3	2	2	F_1	178.793	0.445	4	4	0	F_1	417.380	-0.609
3	2	2	F_2	179.203	0.187	4	4	0	F_2	418.460	-0.812

Observed minus calculated table (040) of \tilde{X}^2B_1

$J K_a K_c$	$F_{1,2}$	Obs.	O-C	$J K_a K_c$	$F_{1,2}$	Obs.	O-C
0 0 0	F_1	5785.567	1.422	3 2 1	F_1	6021.662	1.229
1 0 1	F_1	5806.678	1.216	3 2 1	F_2	6023.012	0.435
1 0 1	F_2	5806.725	1.263	3 3 1	F_1	6141.579	0.363
1 1 1	F_1	5830.312	1.294	3 3 1	F_2	6144.156	-0.047
1 1 1	F_2	5831.195	0.987	3 3 0	F_1	6141.579	0.283
1 1 0	F_1	5836.131	1.727	3 3 0	F_2	6144.292	0.010
1 1 0	F_2	5837.049	1.454	4 0 4	F_1	5983.645	-2.184
2 0 2	F_1	5847.945	0.701	4 0 4	F_2	5983.797	-2.224
2 0 2	F_2	5848.046	0.785	4 1 4	F_1	5992.090	-2.661
2 1 2	F_1	5866.827	0.524	4 1 4	F_2	5995.419	-0.542
2 1 2	F_2	5867.337	-0.157	4 1 3	F_1	6049.329	1.745
2 1 1	F_1	5884.309	1.882	4 1 3	F_2	6049.830	1.006
2 1 1	F_2	5884.905	1.281	4 2 3	F_1	6101.024	-0.242
2 2 1	F_1	5953.463	1.083	4 2 3	F_2	6102.102	-1.400
2 2 1	F_2	5955.335	0.742	4 2 2	F_1	6113.607	1.433
2 2 0	F_1	5954.402	1.210	4 2 2	F_2	6114.735	0.519
2 2 0	F_2	5956.255	0.867	4 3 2	F_1	6228.159	-0.289
3 0 3	F_1	5907.585	-0.363	4 3 2	F_2	6230.259	-1.179
3 0 3	F_2	5907.684	-0.341	4 3 1	F_1	6228.881	-0.110
3 1 3	F_1	5920.935	-0.778	4 3 1	F_2	6230.927	-1.044
3 1 3	F_2	5921.350	-1.560	4 4 1	F_1	6389.273	-0.829
3 1 2	F_1	5955.709	1.925	4 4 1	F_2	6392.409	-1.218
3 1 2	F_2	5956.206	1.211	4 4 0	F_1	6389.303	-0.819
3 2 2	F_1	6017.126	0.599	4 4 0	F_2	6392.437	-1.211
3 2 2	F_2	6018.484	-0.264				

Observed minus calculated table (000) of \tilde{A}^2A_1

$J K_a K_c$	$F_{1,2}$	Obs.	O-C	$J K_a K_c$	$F_{1,2}$	Obs.	O-C
0 0 0	F_1	11122.349	-0.021	3 2 1	F_1	11749.159	-1.534
1 0 1	F_1	11139.922	0.068	3 2 1	F_2	11743.715	2.929
1 0 1	F_2	11139.875	0.021	3 3 1	F_1	12156.677	0.522
1 1 1	F_1	11329.550	-0.411	3 3 1	F_2	12147.746	1.755
1 1 1	F_2	11322.367	2.200	3 3 0	F_1	12156.679	5.445
1 1 0	F_1	11329.084	-2.775	3 3 0	F_2	12147.842	1.848
1 1 0	F_2	11321.778	-0.317	4 0 4	F_1	11297.852	-0.617
2 0 2	F_1	11175.118	0.305	4 0 4	F_2	11297.550	-2.887
2 0 2	F_2	11174.946	0.192	4 1 4	F_1	11495.098	8.908
2 1 2	F_1	11366.047	1.235	4 1 4	F_2	11493.653	17.135
2 1 2	F_2	11362.872	7.840	4 1 3	F_1	11489.625	-15.597
2 1 1	F_1	11364.507	-6.007	4 1 3	F_2	11487.929	-8.101
2 1 1	F_2	11361.229	0.394	4 2 3	F_1	11821.999	-1.732
2 2 1	F_1	11695.305	-0.328	4 2 3	F_2	11818.069	4.157
2 2 1	F_2	11686.892	1.279	4 2 2	F_1	11821.455	-2.481
2 2 0	F_1	11695.257	-0.391	4 2 2	F_2	11817.407	3.163
2 2 0	F_2	11686.841	1.207	4 3 2	F_1	12228.004	-1.380
3 0 3	F_1	11227.630	0.425	4 3 2	F_2	12222.359	3.055
3 0 3	F_2	11227.492	0.397	4 3 1	F_1	12228.534	-0.870
3 1 3	F_1	11421.347	4.418	4 3 1	F_2	12222.328	2.994
3 1 3	F_2	11419.279	12.091	4 4 1	F_1	12692.571	14.735*
3 1 2	F_1	11418.159	-10.184	4 4 1	F_2	12683.410	34.910*
3 1 2	F_2	11415.931	-2.911	4 4 0	F_1	12692.571	46.434*
3 2 2	F_1	11749.351	-1.270	4 4 0	F_2	12683.439	34.868*
3 2 2	F_2	11743.953	3.276				

Results of Analysis I

Large residuals arise for high-lying levels

¹Jungen & Merer, *Molec. Phys.* **40** (1980) 1

²Jensen, Brumm, Kraemer, & Bunker, *J. Molec. Spec.* **171** (1995) 31

Results of Analysis I

Large residuals arise for high-lying levels

Since the effects of the RT-coupling are the largest for such levels, it is suspected that the [Renner interaction](#) should be refined^{1,2}

¹Jungen & Merer, *Molec. Phys.* **40** (1980) 1

²Jensen, Brumm, Kraemer, & Bunker, *J. Molec. Spec.* **171** (1995) 31

Results of Analysis I

Large residuals arise for high-lying levels

Since the effects of the RT-coupling are the largest for such levels, it is suspected that the [Renner interaction](#) should be refined^{1,2}

$$\langle {}^eA_1 | L_z | {}^eB_1 \rangle = i \left[\Lambda - \sum_{i=1}^4 \mathcal{L}^i (1+t)^i \right]$$

¹Jungen & Merer, *Molec. Phys.* **40** (1980) 1

²Jensen, Brumm, Kraemer, & Bunker, *J. Molec. Spec.* **171** (1995) 31

Analysis of the ground and (010) states

Unlike in the previous vibrational states, the effects of the [Renner interaction](#) are very small for the [ground](#) and [\(010\)](#) states which can be treated as isolated states

¹Martin-Drumel, Pirali, and Vervloet, *J. Phys. Chem. A* **118** (2014) 1331

²Coudert, *J. Mol. Spectrosc.* **154** (1992) 427; and Coudert, *ibid.* **165** (1994) 406

Analysis of the ground and (010) states

Unlike in the previous vibrational states, the effects of the [Renner interaction](#) are very small for the [ground](#) and [\(010\)](#) states which can be treated as isolated states

Recent measurements¹ carried out up to $K_a = 13$ showed that an [anomalous centrifugal distortion](#) takes place in these two states

¹Martin-Drumel, Pirali, and Vervloet, *J. Phys. Chem. A* **118** (2014) 1331

²Coudert, *J. Mol. Spectrosc.* **154** (1992) 427; and Coudert, *ibid.* **165** (1994) 406

Analysis of the ground and (010) states

Unlike in the previous vibrational states, the effects of the **Renner interaction** are very small for the **ground** and **(010)** states which can be treated as isolated states

Recent measurements¹ carried out up to $K_a = 13$ showed that an **anomalous centrifugal distortion** takes place in these two states

The **Bending-rotation** approach² previously used to treat the **anomalous centrifugal distortion** of the water molecule is applied to the fitting of data pertaining to these two states

¹Martin-Drumel, Pirali, and Vervloet, *J. Phys. Chem. A* **118** (2014) 1331

²Coudert, *J. Mol. Spectrosc.* **154** (1992) 427; and Coudert, *ibid.* **165** (1994) 406

Effective Hamiltonian

Bending-rotation Hamiltonian¹

$$H_{\text{b-r}} = B_e P_t (1 - t^2) P_t + B_e \left[\frac{N_x^2}{2(1-t)} + \frac{N_y^2}{4} + \frac{N_z^2}{2(1+t)} \right] + V(t) + \dots$$

where B_e is a constant, $t = \cos \tilde{\theta}$, and $V(t)$ is the **bending** potential.

¹Coudert, *J. Mol. Spectrosc.* **165** (1994) 406

²Brown & Sears, *J. Mol. Spectrosc.* **75** (1979) 111

Effective Hamiltonian

Bending-rotation Hamiltonian¹

$$H_{\text{b-r}} = B_e P_t (1 - t^2) P_t + B_e \left[\frac{N_x^2}{2(1-t)} + \frac{N_y^2}{4} + \frac{N_z^2}{2(1+t)} \right] + V(t) + \dots$$

where B_e is a constant, $t = \cos \tilde{\theta}$, and $V(t)$ is the **bending** potential.
Spin-rotation coupling accounted for using²

$$H_{\text{s-r}} = \sum_{\alpha\beta=x,y,z} \epsilon_{\alpha,\beta}(t) N_\alpha S_\beta + \dots$$

where $\epsilon(t)$ is the **spin-rotation** tensor depending on t .

¹Coudert, *J. Mol. Spectrosc.* **165** (1994) 406

²Brown & Sears, *J. Mol. Spectrosc.* **75** (1979) 111

Basis set functions

The bending wavefunctions are expanded using¹

$$\phi_{v_2}^{NkSJ}(t) = (1-t)^{\frac{\alpha}{2}}(1+t)^{\frac{\beta}{2}} \sum_{n=0} a_n^{v_2, NkSJ} P_n^{(\alpha, \beta)}(t)$$

¹Coudert, *J. Mol. Spectrosc.* **154** (1992) 427; and Coudert, *ibid.* **165** (1994) 406

Basis set functions

The bending wavefunctions are expanded using¹

$$\phi_{v_2}^{NkSJ}(t) = (1-t)^{\frac{\alpha}{2}}(1+t)^{\frac{\beta}{2}} \sum_{n=0} a_n^{v_2, NkSJ} P_n^{(\alpha, \beta)}(t)$$

Singularity for the linear configuration taken into account if

$$\alpha = \sqrt{[N(N+1) - k^2]} \quad \text{and} \quad \beta = |k|$$

¹Coudert, *J. Mol. Spectrosc.* **154** (1992) 427; and Coudert, *ibid.* **165** (1994) 406

Data set

Data	State	Reference	N	Max.	
				J	K_a
FIR	Ground	Morino & Kawaguchi ¹	266	13	8
IR	ν_2	Burkholder <i>et al.</i> ²	331	10	5
CD	Ground	Dixon <i>et al.</i> ³	64	18	18
FIR	Ground	Martin-Drumel <i>et al.</i> ⁴	872	25	13
FIR	(010)	Martin-Drumel <i>et al.</i> ⁴	164	19	4
FIR	Ground	<i>This work</i>	210	25	11
All			1907	25	18

¹Morino & Kawaguchi, *J. Mol. Spec.* **182** (1997) 428

²Burkholder, Howard & McKellar, *J. Mol. Spec.* **127** (1988) 415

³Dixon, Irving, Nightingale & Vervloet, *J. Chem. Soc. Faraday Trans.* **87** (1991) 2121

⁴Martin-Drumel, Pirali & Vervloet, *J. Phys. Chem. A* **118** (2014) 1331

Results

Varying 85 parameters the unitless standard deviation was 1.3

¹Morino & Kawaguchi, *J. Mol. Spec.* **182** (1997) 428

²Burkholder, Howard & McKellar, *J. Mol. Spec.* **127** (1988) 415

³Dixon, Irving, Nightingale & Vervloet, *J. Chem. Soc. Faraday Trans.* **87** (1991) 2121

⁴Martin-Drumel, Pirali & Vervloet, *J. Phys. Chem. A* **118** (2014) 1331

Results

Varying 85 parameters the **unitless standard deviation** was 1.3

Data	State	Reference	<i>This work</i>		Ref. [4]	
			<i>N</i>	σ	<i>N</i>	σ
FIR	Ground	Morino & Kawaguchi ¹	266	1.1	269	1.1
IR	ν_2	Burkholder <i>et al.</i> ²	331	1.2	332	0.9
CD	Ground	Dixon <i>et al.</i> ³	64	1.2	52	1.3
FIR	Ground	Martin-Drumel <i>et al.</i> ⁴	872	1.0	835	1.0
FIR	(010)	Martin-Drumel <i>et al.</i> ⁴	164	1.9	165	1.0
FIR	Ground	<i>This work</i>	210	1.7		

¹Morino & Kawaguchi, *J. Mol. Spec.* **182** (1997) 428

²Burkholder, Howard & McKellar, *J. Mol. Spec.* **127** (1988) 415

³Dixon, Irving, Nightingale & Vervloet, *J. Chem. Soc. Faraday Trans.* **87** (1991) 2121

⁴Martin-Drumel, Pirali & Vervloet, *J. Phys. Chem. A* **118** (2014) 1331

High order parameters

Param	Value		Operator
	<i>This work</i>	Ref. [1]	
L_K^0	$+3.0(4) \times 10^{-9}$	-355.7×10^{-9}	N_z^8
L_K^2	$-142.6(170) \times 10^{-9}$		$N_z^8(t - t_e)^2$
M_K^0		$+1.5 \times 10^{-9}$	N_z^{10}
N_K^0		-6.9×10^{-12}	N_z^{12}
O_K^0		$+24.6 \times 10^{-15}$	N_z^{14}
P_K^0		-52.7×10^{-18}	N_z^{16}
Q_K^0		$+49.7 \times 10^{-21}$	N_z^{18}

¹Martin-Drumel, Pirali & Vervloet, *J. Phys. Chem. A* **118** (2014) 1331

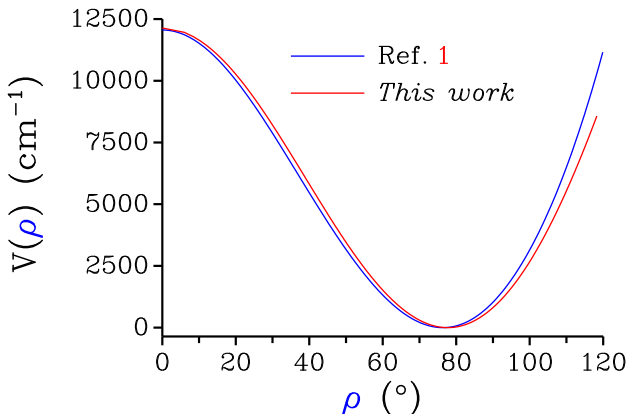
High order parameters

Param	Value		Operator
	<i>This work</i>	Ref. [1]	
L_K^0	$+3.0(4) \times 10^{-9}$	-355.7×10^{-9}	N_z^8
L_K^2	$-142.6(170) \times 10^{-9}$		$N_z^8(t - t_e)^2$
M_K^0		$+1.5 \times 10^{-9}$	N_z^{10}
N_K^0		-6.9×10^{-12}	N_z^{12}
O_K^0		$+24.6 \times 10^{-15}$	N_z^{14}
P_K^0		-52.7×10^{-18}	N_z^{16}
Q_K^0		$+49.7 \times 10^{-21}$	N_z^{18}

The anomalous centrifugal distortion is accounted for

¹Martin-Drumel, Pirali & Vervloet, *J. Phys. Chem. A* **118** (2014) 1331

X^2B_1 bending potential comparison



¹Jensen, Kraemer, & Bunker, *Molec. Phys* **101** (2003) 613

Thank You