

## PARA-ORTHO HYDROGEN CONVERSION; SOLVING A 90-YEAR OLD MYSTERY

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It is well known among spectroscopists that hydrogen has two modifications: para-H<sub>2</sub> and ortho-H<sub>2</sub>. Pure para-H<sub>2</sub> can be produced by leading “normal” H<sub>2</sub>, a 3:1 ortho:para mixture, over a catalyst at low temperature. It is perhaps less well known that para-ortho H<sub>2</sub> conversion is also catalyzed by collisions with paramagnetic molecules, such as O<sub>2</sub>.

Almost ninety years ago Farkas and Sachsse measured the rate coefficient of para-ortho H<sub>2</sub> conversion in gas mixtures with O<sub>2</sub>. [1] In the same year, 1933, it was proposed by Wigner [2] that it is the magnetic dipole-dipole coupling between the electron spin of O<sub>2</sub> and the nuclear spins of the two protons in H<sub>2</sub> that is responsible for the conversion. In asymmetric collisions this coupling makes the two H-nuclei inequivalent and mixes the nuclear spin functions of para- and ortho-H<sub>2</sub>, as well as their rotational states with even and odd  $j$  values. Another mechanism, suggested to be much more effective, was proposed later: the exchange interaction with the open-shell O<sub>2</sub> induces spin density into the electronic wavefunction of H<sub>2</sub>. In most collisions the spin density is different at the two H-nuclei, which makes them inequivalent by different hyperfine interactions through the Fermi contact term.

An important application of para-H<sub>2</sub> is in NMR spectroscopy and its imaging variant, MRI. By adding para-H<sub>2</sub> to the sample the sensitivity of NMR can be increased by four orders of magnitude by a phenomenon called para-hydrogen induced polarization (PHIP). Para-ortho H<sub>2</sub> conversion by O<sub>2</sub> in the gas phase was remeasured in 2014 in view of this application. A detailed and quantitative understanding of the conversion process was still lacking, however.

We theoretically investigated the para-ortho H<sub>2</sub> conversion by collisions with O<sub>2</sub> in a first principles approach. [3] Both mechanisms were taken into account and the corresponding coupling terms were quantitatively evaluated as functions of the geometry of the O<sub>2</sub>-H<sub>2</sub> collision complex by means of *ab initio* electronic structure calculations. Then they were included in nearly exact quantum mechanical coupled-channels scattering calculations for the collisions between O<sub>2</sub> and H<sub>2</sub>, which yielded the para-ortho H<sub>2</sub> conversion cross sections and the rate coefficients for temperatures up to 400 K. The conversion rate and its temperature dependence are in good agreement with the values measured in H<sub>2</sub>-O<sub>2</sub> gas mixtures. The calculations provide detailed insight into the conversion process.

[1] L. Farkas and H. Sachsse, *Z. Phys. Chem. B* **23**, 1 (1933). [2] E. Wigner, *Z. Phys. Chem. B* **23**, 28 (1933). [3] X. Zhang, T. Karman, G. C. Groenenboom, and A. van der Avoird, *Nat. Sci.* (2021); <https://doi.org/10.1002/ntls.10002>.