

## INTERNAL ROTATION OF THE ACETYL METHYL GROUP IN METHYL ALKYL KETONES: THE MICROWAVE SPECTRUM OF OCTAN-2-ONE

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Methyl *n*-alkyl ketones form a class of molecules with interesting internal dynamics in the gas-phase. They contain two methyl groups undergoing internal rotations. To explore the link between structure and barrier heights in ketones, investigations on a series of saturated methyl *n*-alkyl ketones were performed, i.e. pentan-2-one<sup>a</sup> hexan-2-one<sup>b</sup>, heptan-2-one<sup>c</sup> and octan-2-one<sup>d</sup>. The molecular jet Fourier-transform microwave spectrum of octan-2-one was recorded between 2.0 and 40.0 GHz, revealing two conformers, one with C<sub>1</sub> and one with C<sub>s</sub> symmetry. The barriers to internal rotation of the acetyl methyl group were determined to be 233.340(28) cm<sup>-1</sup> and 185.3490(81) cm<sup>-1</sup>, respectively, confirming the link between conformation and barrier height already established for other methyl alkyl ketones. The study combined high level *ab initio* calculations with experimentally derived rotational and torsional parameters using the XIAM<sup>e</sup> and BELGI-C<sub>1</sub> (or BELGI-C<sub>s</sub>)<sup>f</sup> codes. Results from the various fits will be presented. Finally comparisons to molecules in the literature and structural aspects of the conformers generally observed in methyl *n*-alkyl ketones will be reviewed.

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<sup>a</sup>M. Andresen, I. Kleiner, M. Schwell, W. Stahl, H. V. L. Nguyen, *J. Phys. Chem. A* 2018, 122, 7071-7078

<sup>b</sup>M. Andresen, I. Kleiner, M. Schwell, W. Stahl, H. V. L. Nguyen, *ChemPhysChem* 2019, 20, 2063-2073

<sup>c</sup>M. Andresen, I. Kleiner, M. Schwell, W. Stahl, H. V. L. Nguyen, *J. Phys. Chem. A* 2020, 124, 1353-1361.

<sup>d</sup>M. Andresen, D. Schöngen, I. Kleiner, M. Schwell, W. Stahl, H. V. L. Nguyen, *ChemPhysChem* 2020, 21, 2206-2216

<sup>e</sup>H. Hartwig, H. Dreizler, *Z. Naturforsch.* 1996, 51a, 923-932

<sup>f</sup>I. Kleiner, J. T. Hougen, *J. Chem. Phys.* 2003, 119, 5505-5509.