

Valence-hole electron configurations: A new global electronic structure paradigm for C₂ (and possibly beyond)

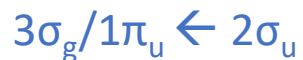
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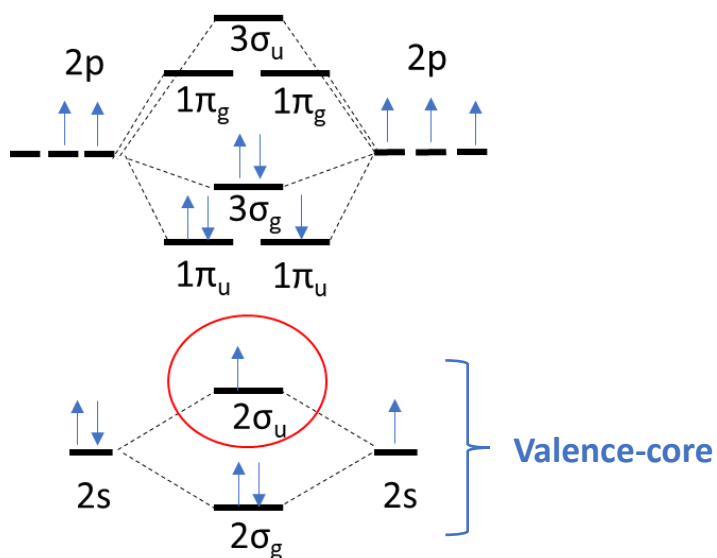
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Chemical intuition!!!

Valence-hole state:



- Nominal bond order of 3
- At large R, correlates to $2s^2 2p^2 + \underline{2s^1 2p^3}$



- The electronic structures of low-lying $^1\Pi_g$ and $^3\Pi_g$ states are systematically disrupted due to curve-crossings with the deeply-bound, valence-hole states
- The effects of the valence-hole-induced curve-crossings are gloriously sampled in the molecular constants derived from the laboratory spectra of C₂

