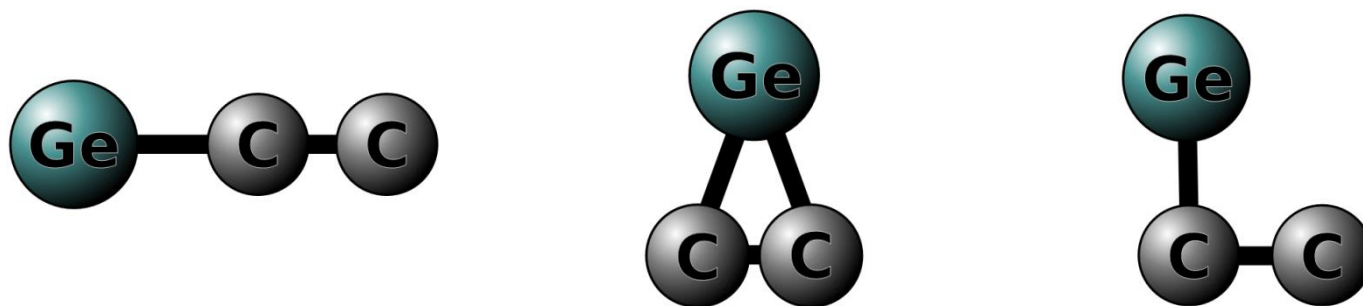


Germanium Dicarbide: Evidence for a T-shaped Ground State Structure



O. Zingsheim, M.A. Martin-Drumel, S. Thorwirth,
S. Schlemmer, C.A. Gottlieb, J. Gauss, and M.C. McCarthy

GeC₂ – A vivid example

- Since the late 1950s:
Discussion about the structure of GeC₂
- Isovalent to SiC₂ (C_{2v} structure¹)
- Metal Dicarbides MC₂ are detected in space
(e.g. C₂O² and SiC₂³)
- Carbon-rich Group 14 materials: Attractive for
nanoelectronics and material science

[1] Michalopoulos et al., *J. Chem. Phys.* **80** (1984)

[2] Ohishi et al., *Astrophys. J.* **380** (1991)

[3] Thaddeus et al., *Astrophys. J.* **283** (1984)

GeC₂ – A vivid example

- Since the late 1950s:
Discussion about the structure of GeC₂
- Isovalent to SiC₂ (C_{2v} structure¹)
- Metal Dicarbides MC₂ are detected in space (e.g. C₂O² and SiC₂³)
- Carbon-rich Group 14 materials: Attractive for nanoelectronics and material science

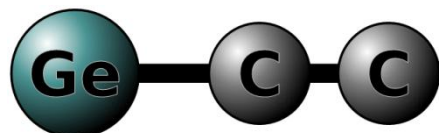
[1] Michalopoulos et al., *J. Chem. Phys.* **80** (1984)

[2] Ohishi et al., *Astrophys. J.* **380** (1991)

[3] Thaddeus et al., *Astrophys. J.* **283** (1984)

GeC₂ – Structural debate

linear



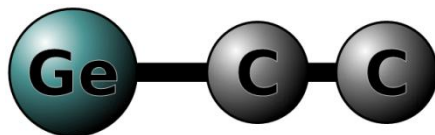
GeC₂ – Structural debate

THE JOURNAL OF CHEMICAL PHYSICS

VOLUME 30, NUMBER 1

JANUARY, 1959

linear



Mass Spectrometric Study of Inter-Group IVB Molecules*

J. DROWART,[†] G. DE MARIA,[‡] A. J. H. BOERBOOM,[§] AND M. G. INGRAM

Department of Physics, University of Chicago, Chicago, Illinois

(Received August 8, 1958)

The systems germanium-graphite and germanium-silicon-carbon have been studied. Several gaseous molecules containing Ge, Si, and/or C were observed and their dissociation energies determined from measured partial pressures. A general picture of the dissociation energies of di-, tri-, and tetraatomic intergroup IVB molecules is deduced from these measurements.

Incidental to this study the vapor pressure of pure Si was determined; the result confirms the heat of sublimation obtained previously.

THEO
CHEM

Journal of Molecular Structure (Theochem) 629 (2003) 295–302

www.elsevier.com/locate/theochem



ELSEVIER

Structure and stability of Ge_nC_{m-n} clusters

Şenay Katırcıoğlu*

Department of Physics, Middle East Technical University, Ankara 06531, Turkey

Received 21 January 2003; revised 3 March 2003; accepted 6 March 2003

Abstract

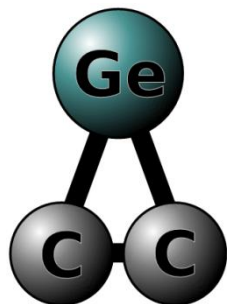
Density functional theory method has been used to predict the structural and energetic properties of Ge_nC_{m-n} ($n = 1, m = 3-14$; $n = 2, m = 4-15$) isomers. The optimized stable structures have been found to be linear(chain) for all Ge_nC_{m-n} ($n = 1, m = 3-9$; $n = 2, m = 4-10$) clusters and planar(ring) for all Ge_nC_{m-n} ($n = 1, m = 10-14$; $n = 2, m = 11-15$) isomers. It has been also found that the Ge_nC_{m-n} ($n = 1, m = 3-14$; $n = 2, m = 4-15$) clusters with odd m are more stable than those with even m .

© 2003 Elsevier B.V. All rights reserved.

Keywords: Ge_nC_{m-n} clusters; Stability; Dipole moment; Density functional theory; B3LYP

GeC₂ – Structural debate

T-shaped



GeC₂ – Structural debate

PHYSICAL REVIEW B, VOLUME 64, 195312

Structural and electronic properties of semiconductor binary microclusters A_mB_n ($A, B = \text{Si, Ge, C}$): A B3LYP-DFT study

Si-Dian Li,^{1,2} Zhi-Guang Zhao,² Xiu-Feng Zhao,² Hai-Shun Wu,³ and Zhi-Hao Jin¹

¹School of Materials Science and Engineering, Xian Jiaotong University, Xian 710049, People's Republic of China

²Department of Chemistry, Xinzhou Teachers' University, Xinzhou 034000, Shanxi, People's Republic of China

³Institute of Materials Chemistry, Shanxi Normal University, Linfen 041000, Shanxi, People's Republic of China

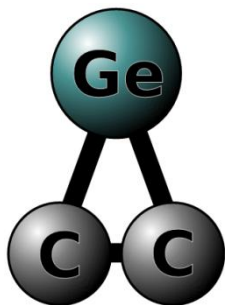
(Received 26 April 2001; published 15 October 2001)

Structural and electronic properties of semiconductor binary microclusters A_mB_n ($A, B = \text{Si, Ge, C}$) have been investigated using the B3LYP-DFT method in the range $s = m + n \leq 10$. Full structural optimization and frequency analyses are performed with the basis of 6-311G(3df). Geometries of A_mB_n binary clusters are found to follow similar structural patterns with lower symmetries when compared with corresponding elemental Si₂ and Ge₂ in this size range. The optimized structures have either singlet or triplet ground states, depending on specific cluster size, cluster composition, and configurations. Similar to the ionization potentials of Ge₂ clusters in the same size range, the calculated vertical ionization potentials of Si_mGe_n vary with an even-odd alternation in the range of $s = 2 - 7$, a global minimum at $s = 8$ ($C_5\text{Si}_4\text{Ge}_4$) and an obvious recovery at $s = 9$ ($C_{2v}\text{Si}_5\text{Ge}_4$) and $s = 10$ ($C_{3v}\text{Si}_6\text{Ge}_4$). Both Si₄Ge₆ and Si₆Ge₄ are predicted to be species with high stabilities and possible to be produced experimentally.

DOI: 10.1103/PhysRevB.64.195312

PACS number(s): 73.22.-f, 61.46.+w

T-shaped



15294

J. Phys. Chem. **1995**, 99, 15294–15297

Atomization Enthalpies and Enthalpies of Formation of the Molecules GeC₂, Ge₂C, Ge₂C₂, and Ge₃C from Mass Spectrometric Equilibrium Experiments

Richard W. Schmude, Jr.,[†] and Karl A. Gingerich*

Department of Chemistry, Texas A&M University, College Station, Texas 77843

Joseph E. Kingcade

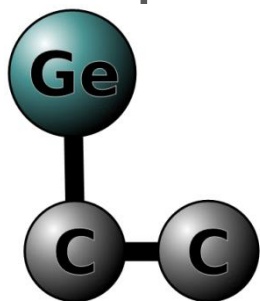
Blinn College, Division of Natural Science, Brenham, Texas 77833

Received: June 23, 1995[®]

The Knudsen effusion mass spectrometric method was used in measuring the partial pressures of GeC₂, Ge₂C, Ge₃C, and Ge₂C₂ above liquid germanium which was contained in a graphite Knudsen cell. These partial pressures have been combined with calculated thermal functions to determine the atomization enthalpies, $\Delta H_{a,0}$, (in kJ mol⁻¹): 1197 ± 10 (GeC₂); 902 ± 12 (Ge₂C); 1247 ± 22 (Ge₃C); and 1535 ± 25 (Ge₂C₂). The enthalpies of formation, $\Delta H_{f,0}$, (in kJ mol⁻¹) are 597 ± 10 (GeC₂); 552 ± 12 (Ge₂C); 579 ± 22 (Ge₃C); and 630 ± 25 (Ge₂C₂).

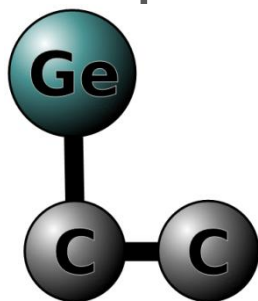
GeC₂ – Structural debate

L-shaped



GeC₂ – Structural debate

L-shaped



THE JOURNAL OF CHEMICAL PHYSICS 133, 124306 (2010)

Structure and bonding in third-row main group dicarbides C₂X (X=K–Br)

Víctor M. Rayón, Pilar Redondo, Carmen Barrientos, and Antonio Largo^{a)}
*Departamento de Química Física y Química Inorgánica, Computational Chemistry Group,
 Facultad de Ciencias, Universidad de Valladolid, 47011 Valladolid, Spain*

(Received 28 April 2010; accepted 8 July 2010; published online 23 September 2010)

THE JOURNAL OF CHEMICAL PHYSICS 123, 234309 (2005)

Thermodynamic properties of germanium/carbon microclusters

Paweł Wielgus and Szczepan Roszak
*Institute of Physical and Theoretical Chemistry, Wrocław University of Technology, Wybrzeże Wyspiańskiego
 27, 50-370 Wrocław, Poland and Computational Center for Molecular Structure and Interactions,
 Jackson State University, Jackson, Mississippi 39217*

Devashis Majumdar and Jerzy Leszczynski^{a)}
*Computational Center for Molecular Structure and Interactions, Jackson State University, Jackson,
 Mississippi 39217*

(Received 9 September 2005; accepted 20 October 2005; published online 22 December 2005)

Physica E 74 (2015) 175–192



Contents lists available at ScienceDirect

Physica E

journal homepage: www.elsevier.com/locate/physa



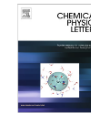
Chemical Physics Letters 659 (2016) 216–220



Contents lists available at ScienceDirect

Chemical Physics Letters

journal homepage: www.elsevier.com/locate/cplett



Structural, electronic and vibrational properties of Ge_xC_y (x+y=2–5) nanoclusters: A B3LYP-DFT study

Sohini Goswami, Sushmita Saha, R.K. Yadav^{*}

Department of Physics, National Institute of Technology Agartala, Tripura 799046, India



Research paper

Regioselectivity of third-row main group dicarbides, C₂X (X = K–Br) for CO interaction: Fukui function and topological analyses

Saroj K. Parida^a, Sridhar Sahu^{a,*}, Sagar Sharma^{b,1}

^a*Department of Applied Physics, Indian School of Mines, Dhanbad, Jharkhand 826004, India*

^b*Department of Organic Chemistry, Weizmann Institute of Science, Rehovot 76100, Israel*



An L-shaped equilibrium geometry for germanium dicarbide (GeC_2)? Interesting effects of zero-point vibration, scalar relativity, and core–valence correlation

Levent Sari

Center for Computational Quantum Chemistry, University of Georgia, Athens, Georgia 30602-2525

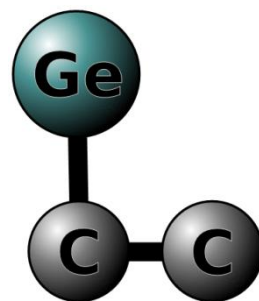
Kirk A. Peterson

Department of Chemistry, Washington State University, Richland, Washington 99352

Yukio Yamaguchi and Henry F. Schaefer III^{a)}

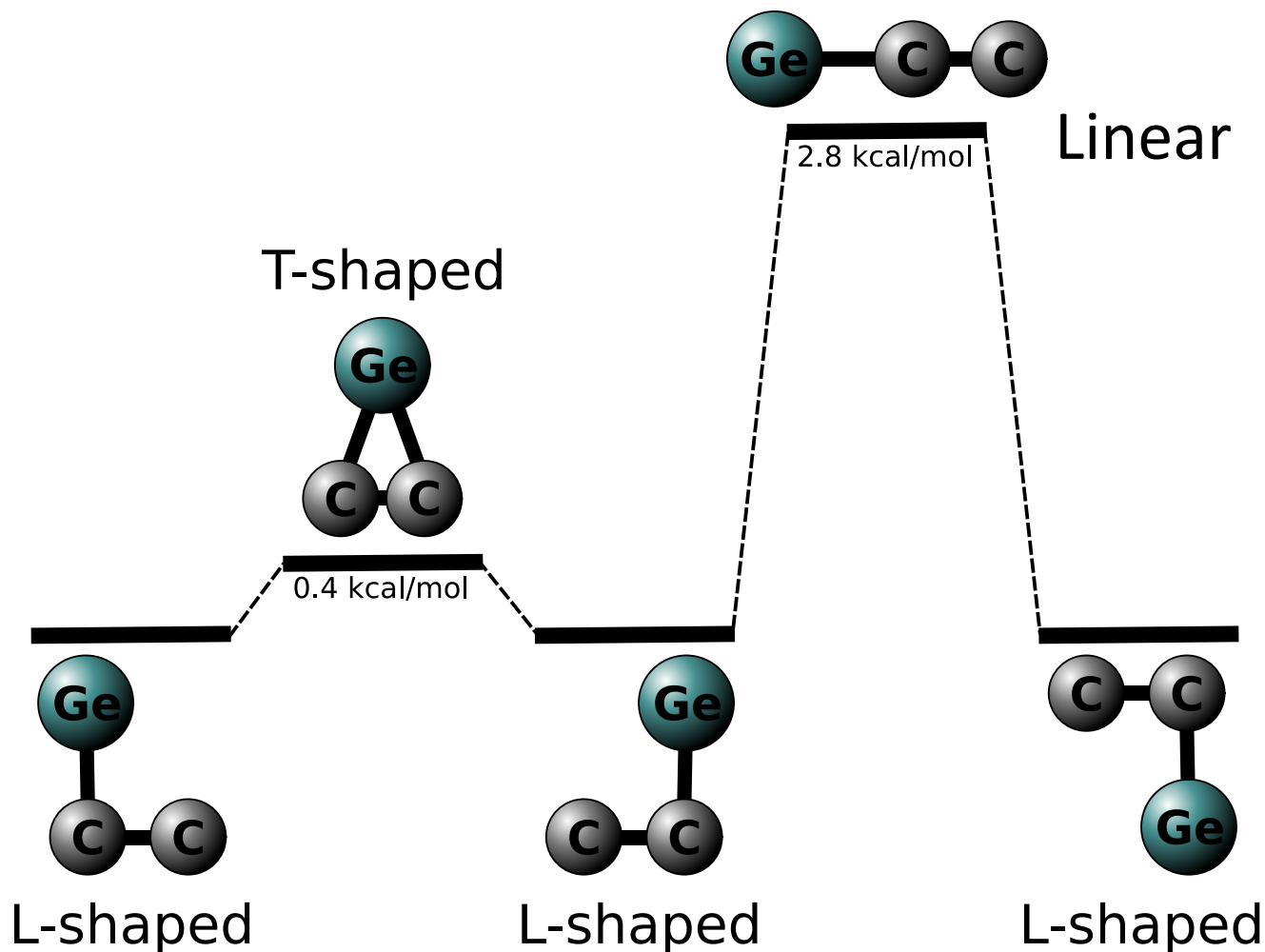
Center for Computational Quantum Chemistry, University of Georgia, Athens, Georgia 30602-2525

(Received 11 July 2002; accepted 12 September 2002)

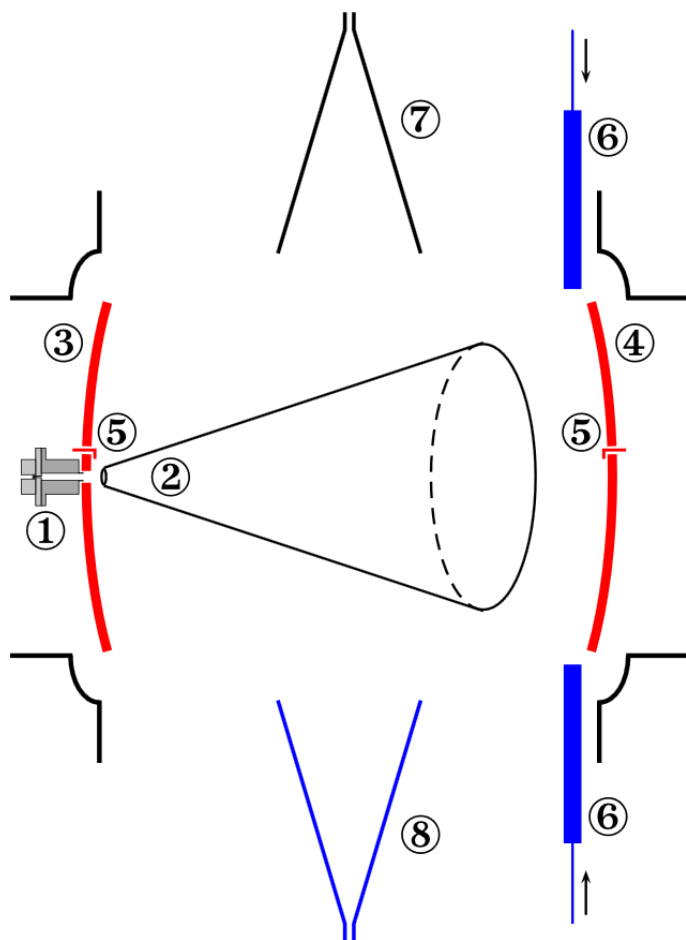


The ground state potential energy surface of the GeC_2 molecule has been investigated at highly correlated coupled cluster levels of theory. Large basis sets including diffuse functions and functions to describe core correlation effects were employed in order to predict the true equilibrium geometry for GeC_2 . Like the much-studied valence isoelectronic SiC_2 , the linear ($^1\Sigma^+$), L-shaped ($^1A'$), and T-shaped structures (1A_1) must be investigated. The L-shaped C_∞ geometry is found to have real harmonic vibrational frequencies along every internal coordinate, and the linear stationary point has an imaginary vibrational frequency along the bending mode at every level of theory employed. The T-shaped geometry is found to have an imaginary vibrational frequency along the asymmetric stretching mode. At the coupled cluster with single and double excitations and perturbative triple excitations [CCSD(T)]/correlation consistent polarized valence quadrupole- ζ (cc-pVQZ) level, the nonrelativistic classical relative energies of the T-shaped and linear structures with respect to the L-shaped minimum are 0.1 and 2.8 kcal/mol, respectively. Including zero-point vibrational energy, scalar relativistic, and core-valence corrections, the T-L energy separation is shifted to 0.4 kcal/mol and the relative energy between the L-shaped and linear structures is still 2.8 kcal/mol. All nonrelativistic and relativistic computations predict that the L-shaped ($^1A'$) structure is most favored for the ground state. The linear structure is predicted to be a transition state, as the case of SiC_2 . © 2002 American Institute of Physics. [DOI: 10.1063/1.1518966]

GeC₂ – Structural debate



Chirped-Pulse Spectrometer



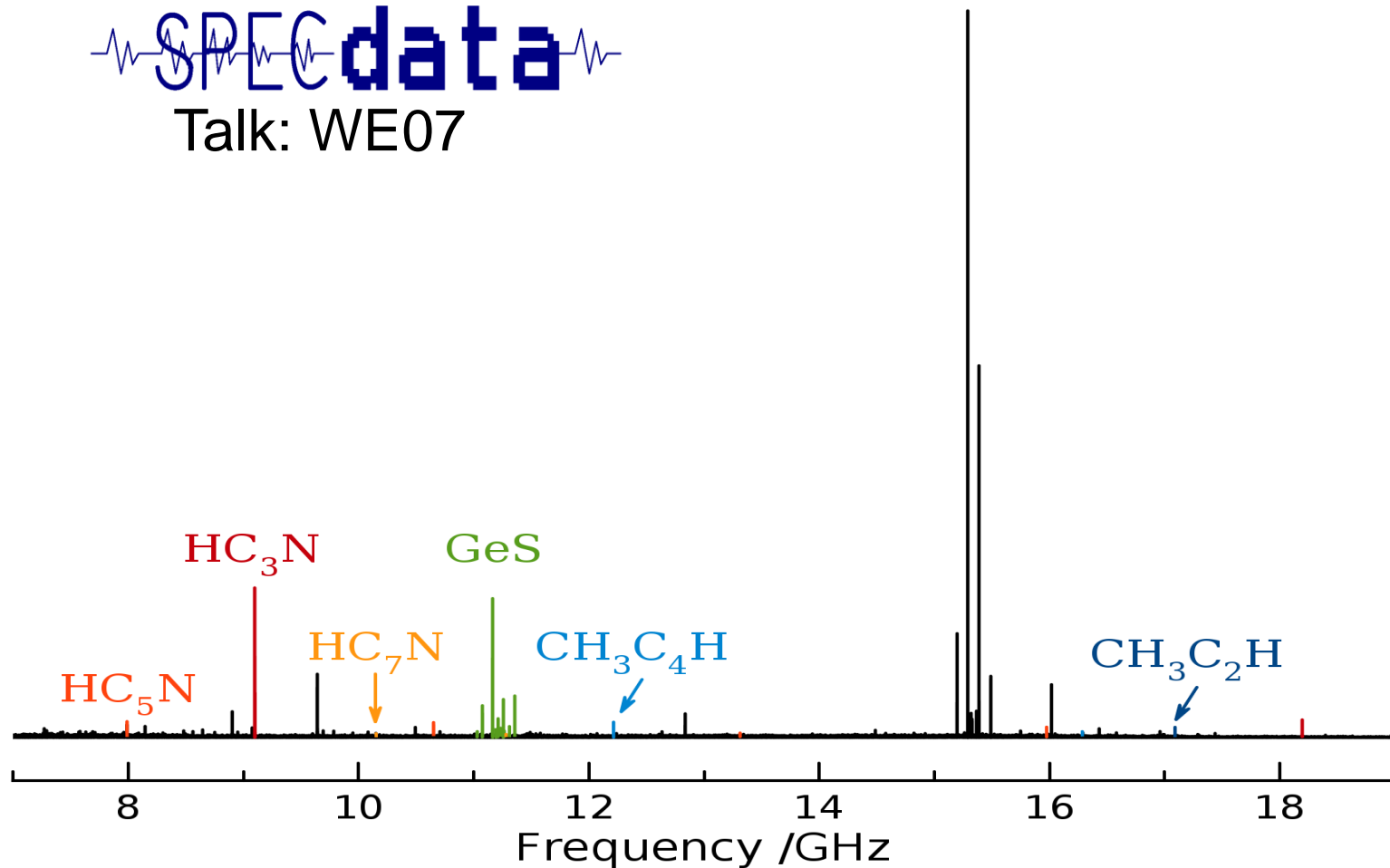
- ① Pulsed nozzle
- ② Supersonic expansion
- ③ Fixed mirror
- ④ Movable mirror
- ⑤ Microwave emitter/receiver
- ⑥ Microwave absorbers
- ⑦ Horn antenna
- ⑧ Horn antenna

Crabtree et al, *J. Chem. Phys.* **144** (2016)

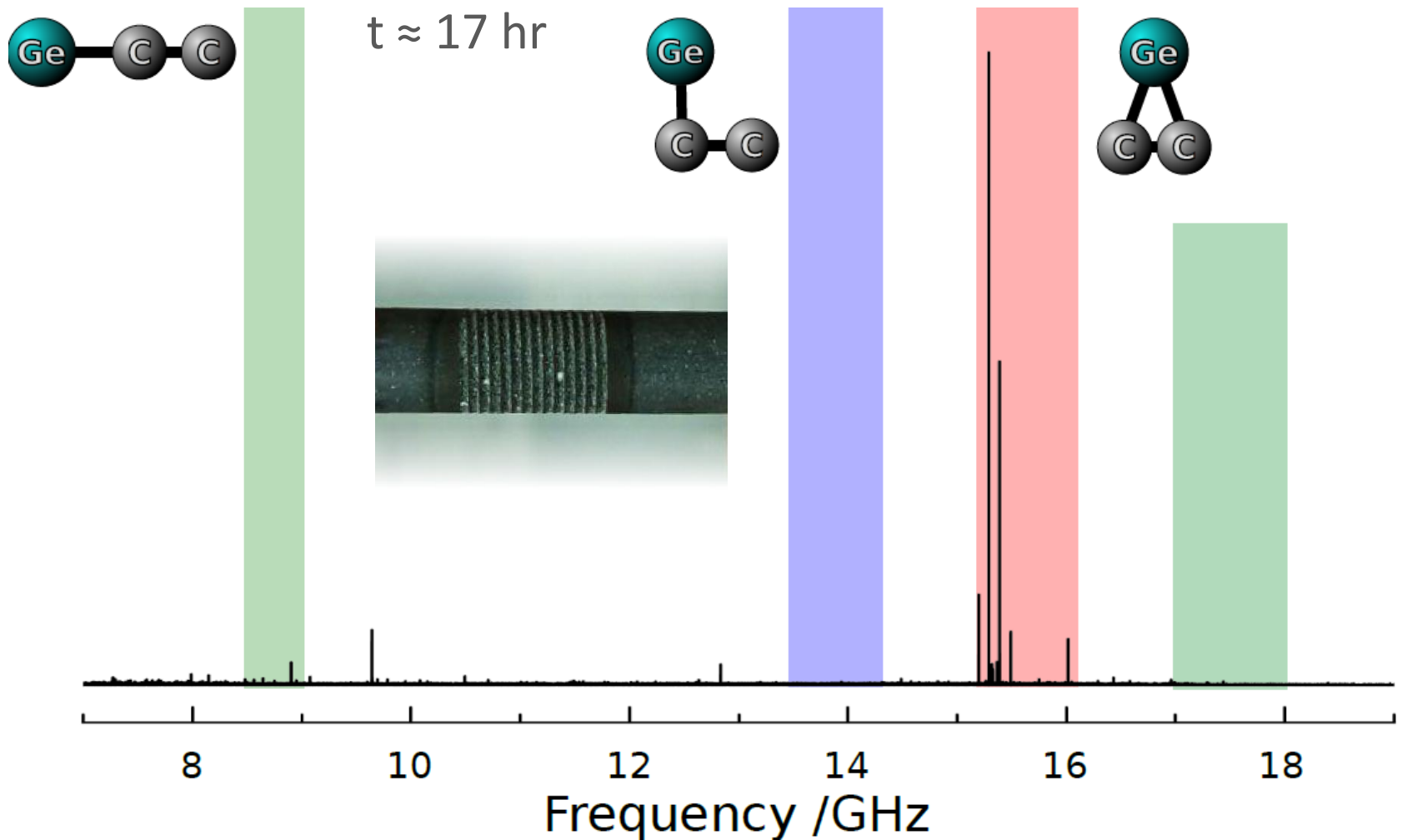
Broadband measurement

SPECdata

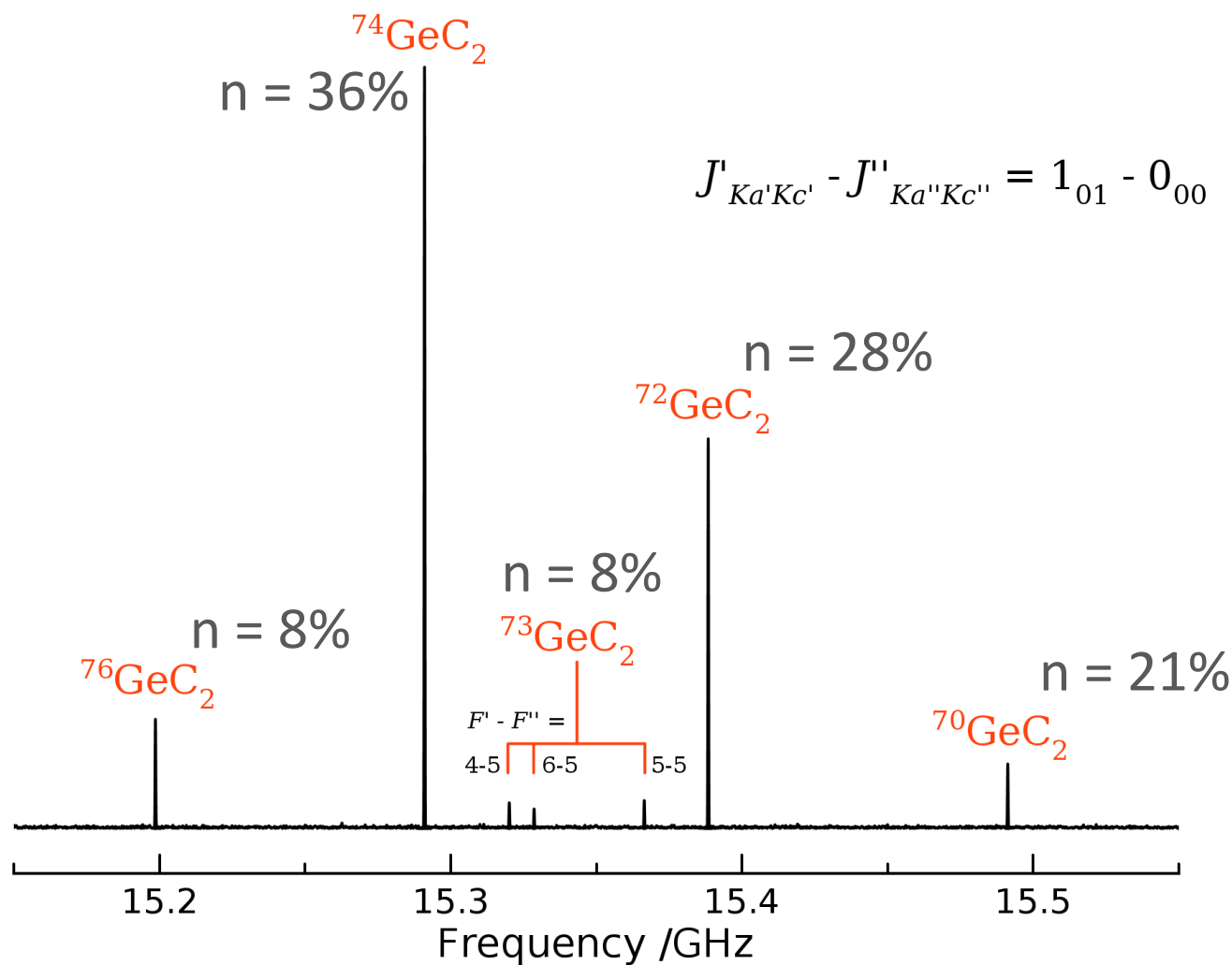
Talk: WE07



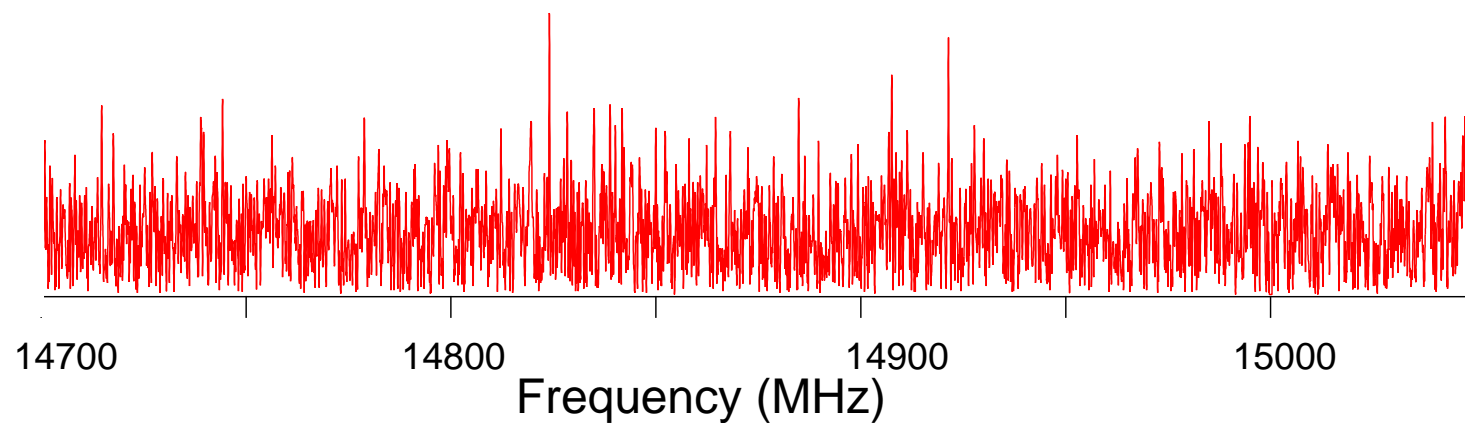
Broadband measurement



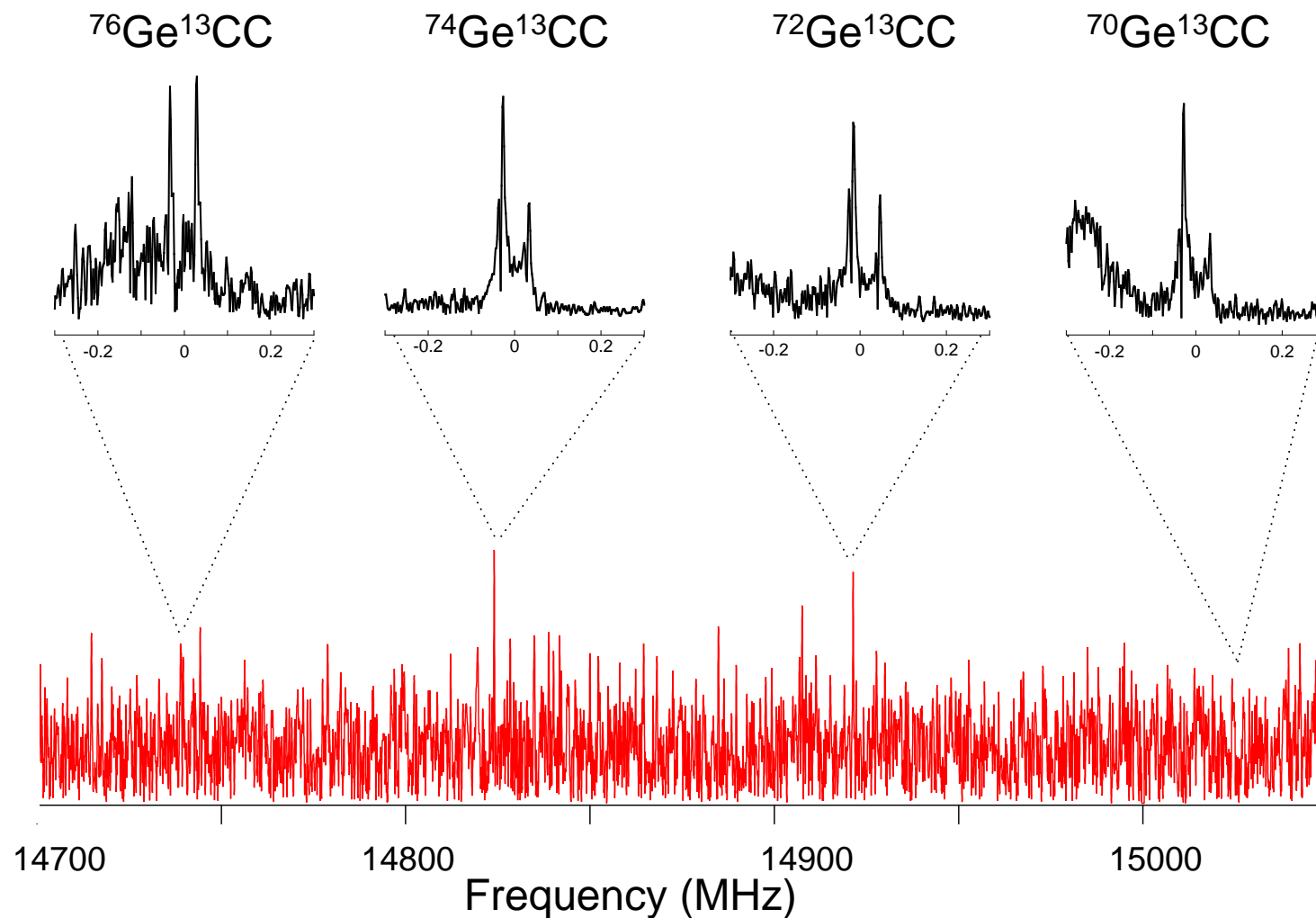
Zoom



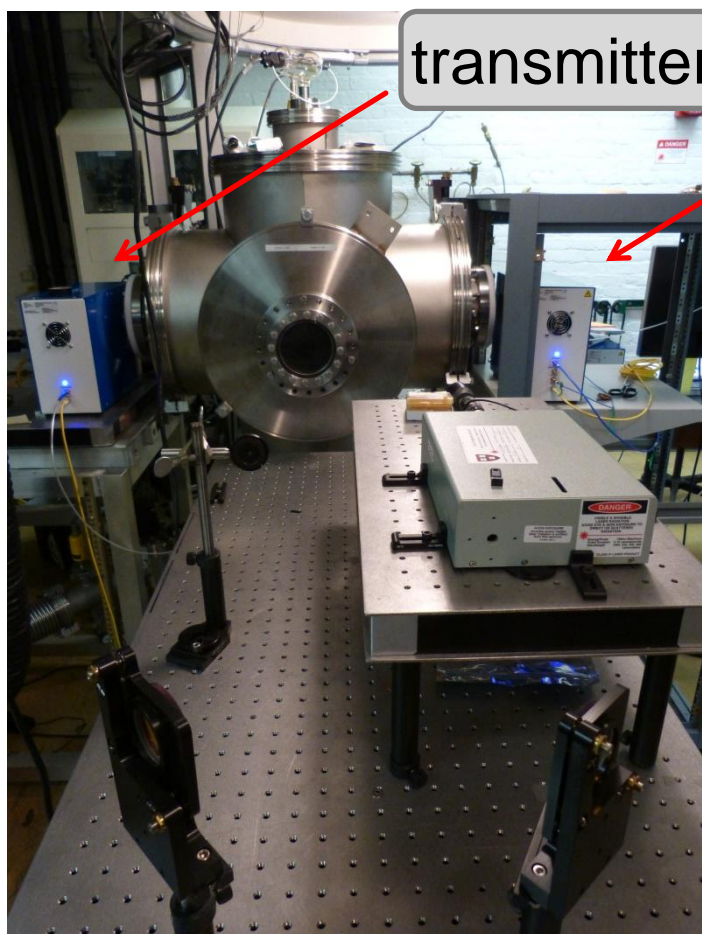
Wealth of Isotopologues



Wealth of Isotopologues



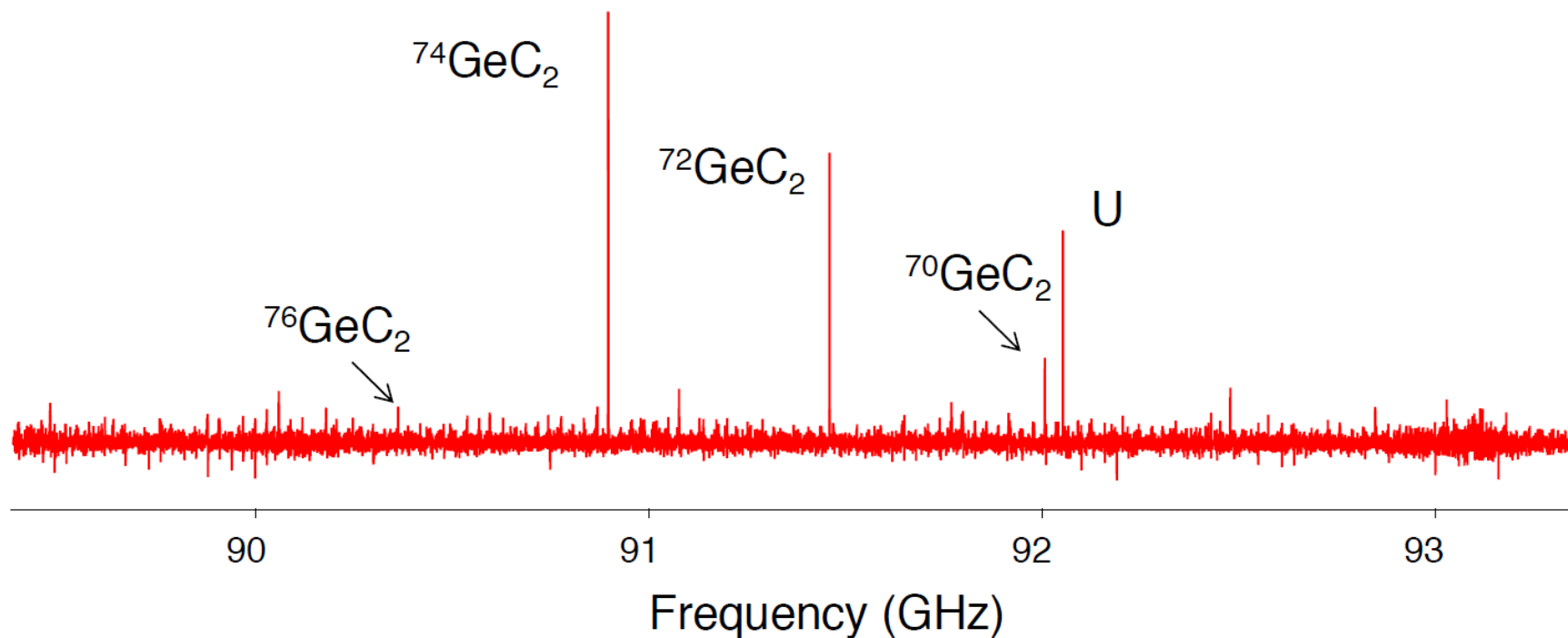
W-Band spectrometer



- Commercial chirped spectrometer
- 75-115 GHz



W-Band measurement

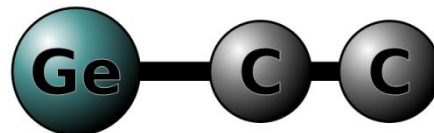


Rotational parameters

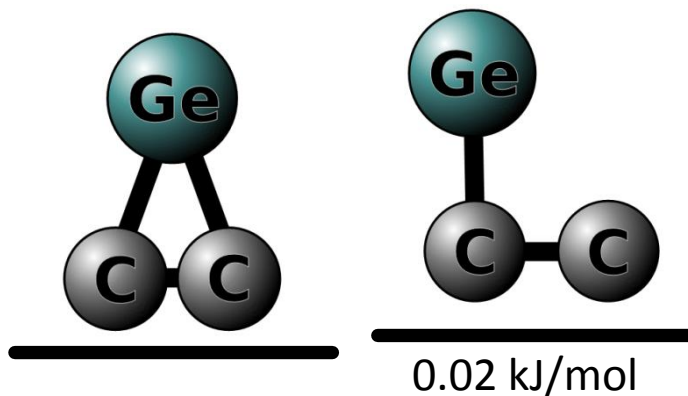
	Constant	$^{12}\text{C}_2$	$^{12}\text{C}^{13}\text{C}$	$^{13}\text{C}_2$
^{70}Ge	A_0	[53431.4]	51397.0(201)	49297.4(221)
	$(B_0 + C_0)/2$	7745.5872(50)	7512.07635(120)	7294.79366(198)
	$(B_0 - C_0)/4$	317.74(39)	306.80032(85)	301.07628(85)
^{72}Ge	A_0	[53431.4]	51381.8(207)	49309.0(228)
	$(B_0 + C_0)/2$	7694.1462(52)	7460.67187(120)	7243.44571(198)
	$(B_0 - C_0)/4$	313.80(40)	302.63221(85)	296.87040(85)
^{73}Ge	A_0	[53431.4]	51377.9(109)	—
	$(B_0 + C_0)/2$	7669.41276(58)	7435.95145(144)	—
	$(B_0 - C_0)/4$	311.744(60)	300.6374(48)	—
^{74}Ge	A_0	[53431.4]	51382.3(213)	49308.4(235)
	$(B_0 + C_0)/2$	7645.4226(53)	7411.97649(120)	7194.80702(198)
	$(B_0 - C_0)/4$	309.72(42)	298.70926(85)	292.91178(85)
^{76}Ge	A_0	[53431.4]	51343.6(90)	[49308.6]
	$(B_0 + C_0)/2$	7599.2042(55)	7365.78453(95)	7148.66602(112)
	$(B_0 - C_0)/4$	305.93(44)	295.01088(88)	289.18212(88)

Structural parameters

CCSD(T)/cc-pwCVQZ - level

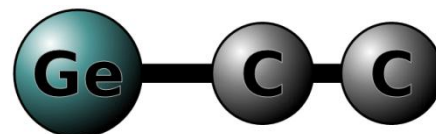


14.74 kJ/mol



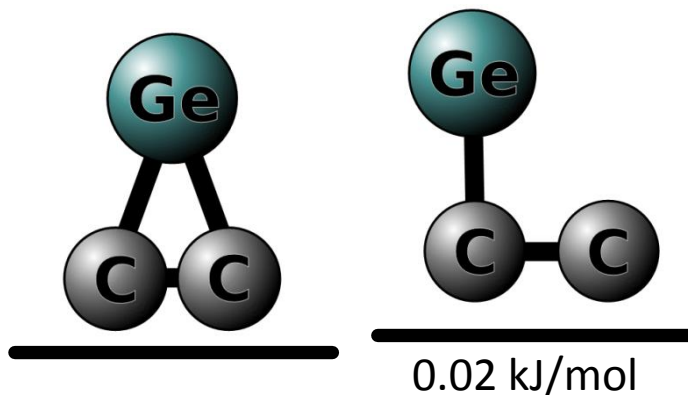
Structural parameters

CCSD(T)/cc-pwCVQZ - level



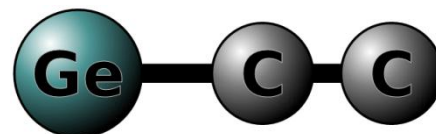
14.74 kJ/mol

	Experimental	T-shaped	L-shaped	Linear
$r_{\text{Ge-C}} / \text{\AA}$	1.952(1)	1.9296	1.8184	1.7717
$r_{\text{C-C}} / \text{\AA}$	1.294	1.2750	1.2810	1.2836
Apex angle /°	38.7(2)	38.58	86.29	



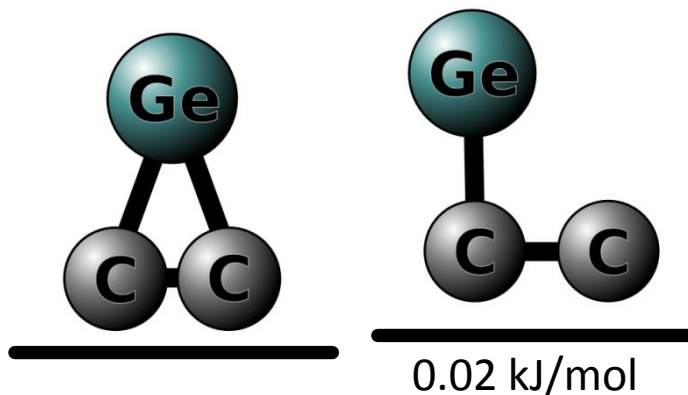
Structural parameters

CCSD(T)/cc-pwCVQZ - level



14.74 kJ/mol

	Experimental	T-shaped	L-shaped	Linear
$r_{\text{Ge-C}} / \text{\AA}$	1.952(1)	1.9296	1.8184	1.7717
$r_{\text{C-C}} / \text{\AA}$	1.294	1.2750	1.2810	1.2836
Apex angle / °	38.7(2)	38.58	86.29	



Acknowledgments

Thank you for your attention!



AST-1615847



➤ Justin Neill and Matt Muckle

