NEAR-INFRARED SPECTROSCOPY OF SMALL PROTONATED WATER CLUSTERS

J. PHILIPP WAGNER, Department of Chemistry, University of Georgia, Athens, GA, USA; DAVID C Mc-DONALD, Chemistry, University of Georgia, Athens, GA, USA; ANNE B McCOY, Department of Chemistry, University of Washington, Seattle, WA, USA; MICHAEL A DUNCAN, Department of Chemistry, University of Georgia, Athens, GA, USA.

Small protonated water clusters and their argon tagged analogues of the general formula $H^+(H_2O)_nAr_m$ have been generated in a pulsed electric discharge source. Clusters containing n=1-8 water molecules were mass-selected and their absorptions in the near-infrared were probed with a tunable Nd: YAG pumped OPA/OPA laser system in the region from $4850-7350~\rm cm^{-1}$. A doublet corresponding to overtones of the free O-H stretches of the external waters was observed around $7200~\rm cm^{-1}$ that was continuously decreasing in intensity with increasing cluster size. Broad, mostly featureless absorptions were found around $5300~\rm cm^{-1}$ associated with stretch/bend combinations and with the hydrogen bonded waters in the core of the clusters. Vibrational assignments were substantiated by comparison to anharmonic frequency computations via second-order vibrational perturbation theory (VPT2) at the MP2/aug-cc-pVTZ level of theory.