

# CI-LOSS DYNAMICS OF VINYL CHLORIDE CATION IN $B^2A''$ STATE: ROLE OF $C^2A'$ STATE

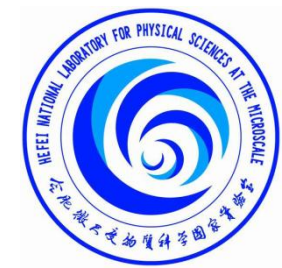
— Threshold **P**hoto**e**lectron-**P**hoto**i**on **C**oincidence **V**elocity **M**ap **I**maging

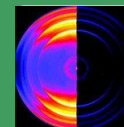
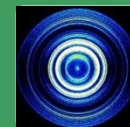
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Microscale and Department of Chemical Physics, University of  
Science and Technology of China*



2017.06.20





## ◆ Spectroscopy and dynamics of molecular ions

- ◆ open-shell system

- ◆ accurate parameters

  - e.g. **ionization potential (IP), appearance potential (AP)**

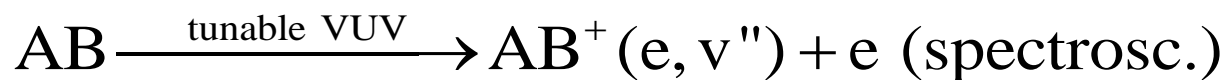
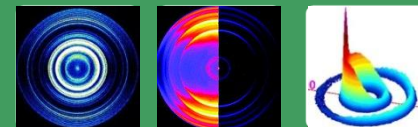
  - bond dissociation energy (BDE)**

- ◆ molecular interactions

  - e.g. **conical intersection, internal conversion, Jahn-Teller**

- ◆ benchmark for high-level theoretical calculation

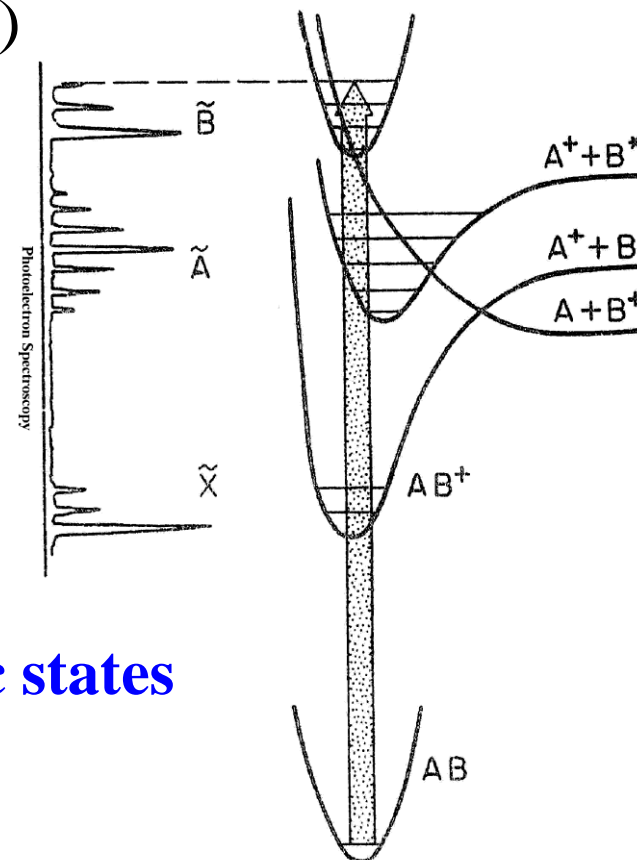
# VUV photoionization or DPI



Energy conservation:

$$E_{AB^+} = E_{AB} + h\nu - \text{IP}(AB) - E_{el}$$

●  $E_{el} = 0$ , **TPEPICO**



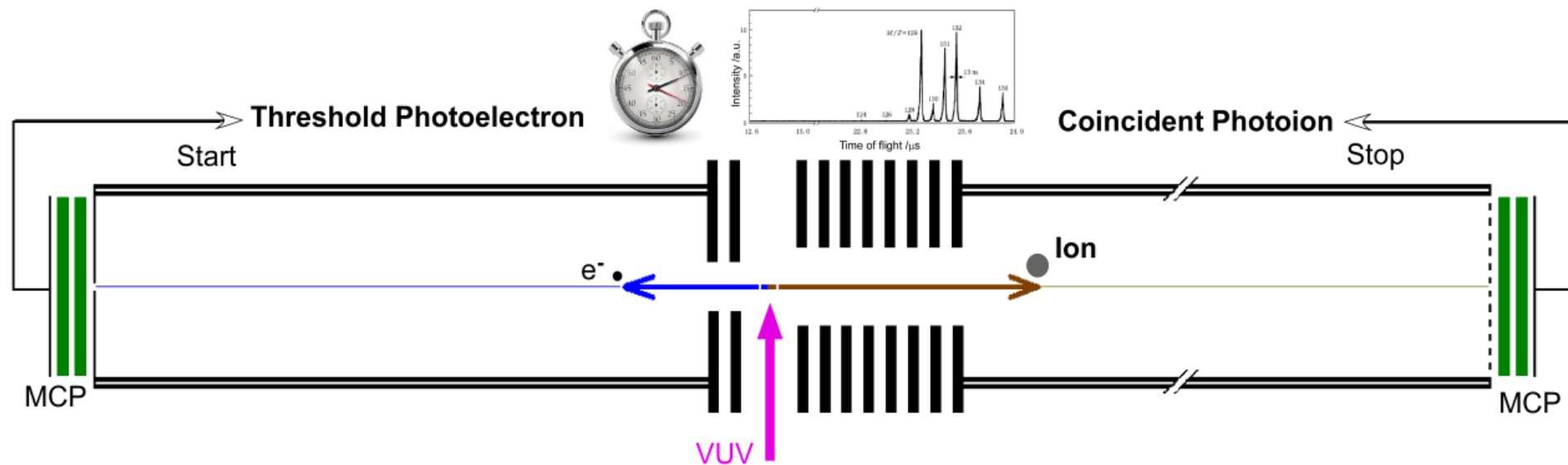
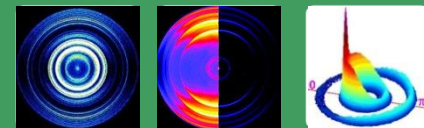
**General method, suitable for all electronic states**

In a dissociative photoionization event:

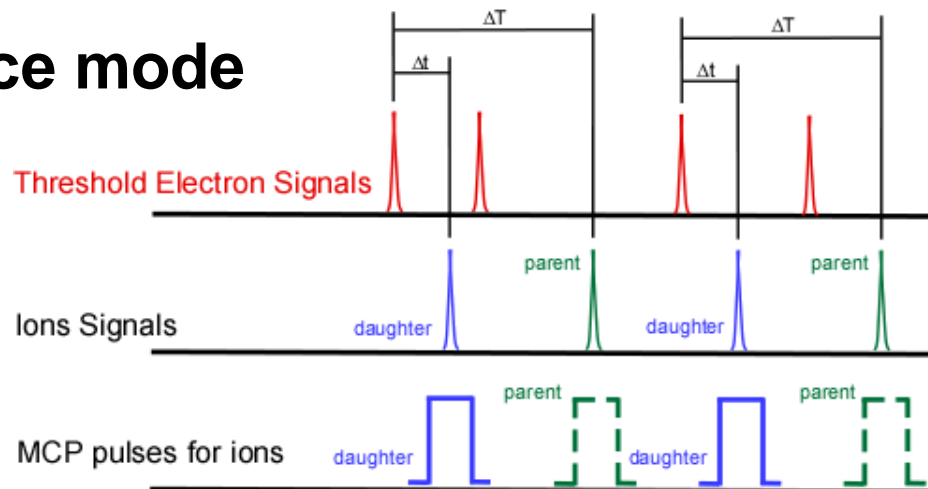
$$E_{in}(A^+) + E_{in}(B) + E_T = E_{AB} + h\nu - \text{IP}(AB) - E_{el} - \underline{\text{BDE}(AB^+)}$$

$$E_T = E_T(A^+) + E_T(B) : \text{kinetic energy released distribution (KERD)}$$

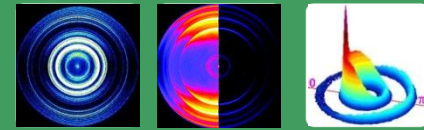
# TPEPICO principle



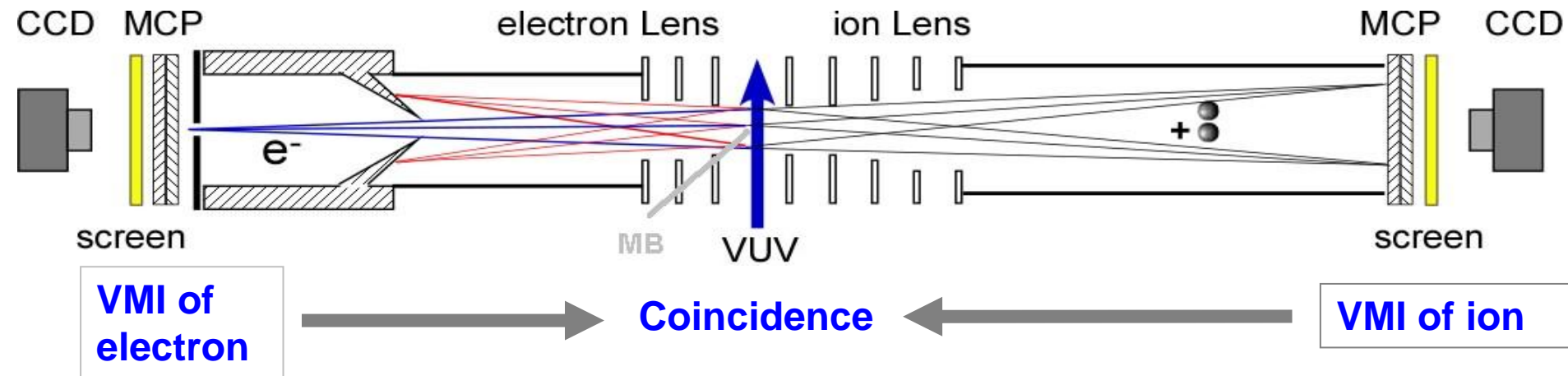
## Time coincidence mode



# New TPEPICO design at Hefei

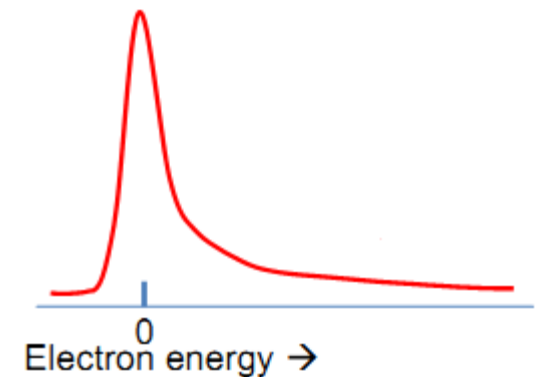


## TPEPICO + double Velocity Imaging

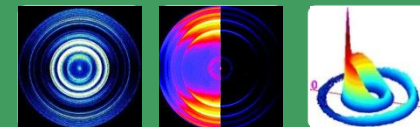


### Key points:

- ◆ Hot electron tail in TPES;
- ◆ False coincidence measurement

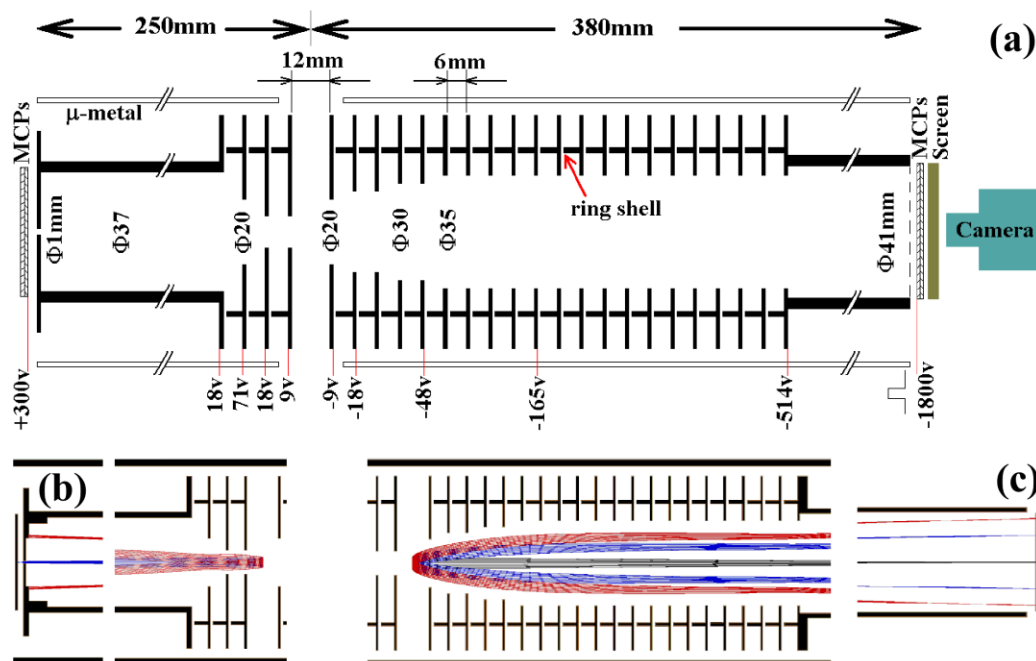


# Key points in TPEPICO VMI



## decelerated ion optics

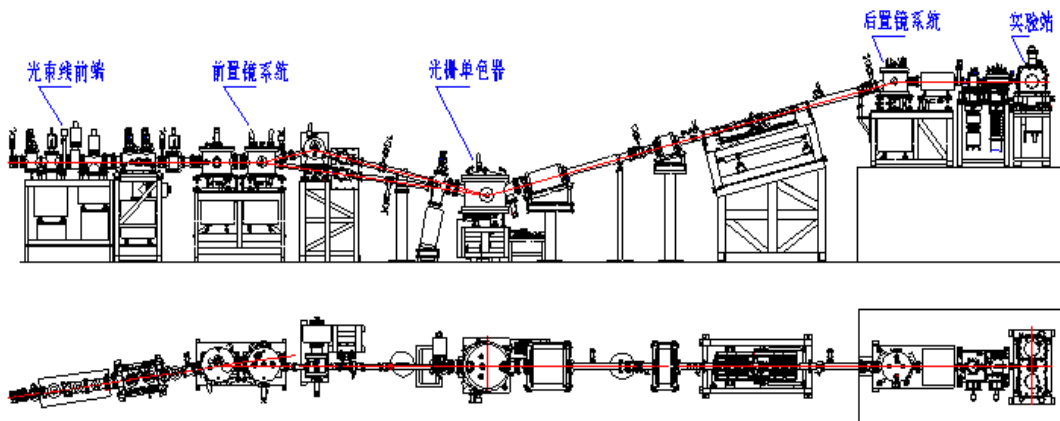
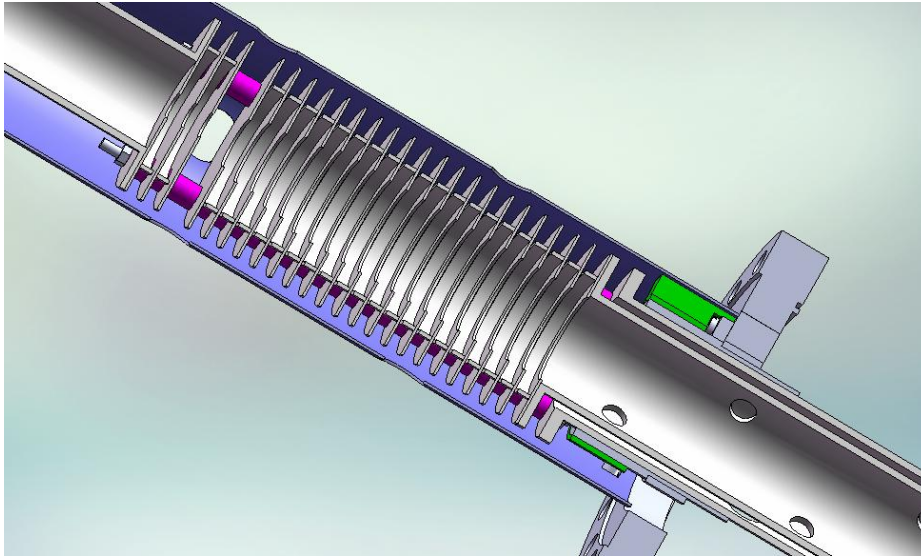
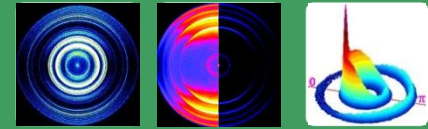
- Reduce the ratio of hot electrons falsely detected in coincidence measurement



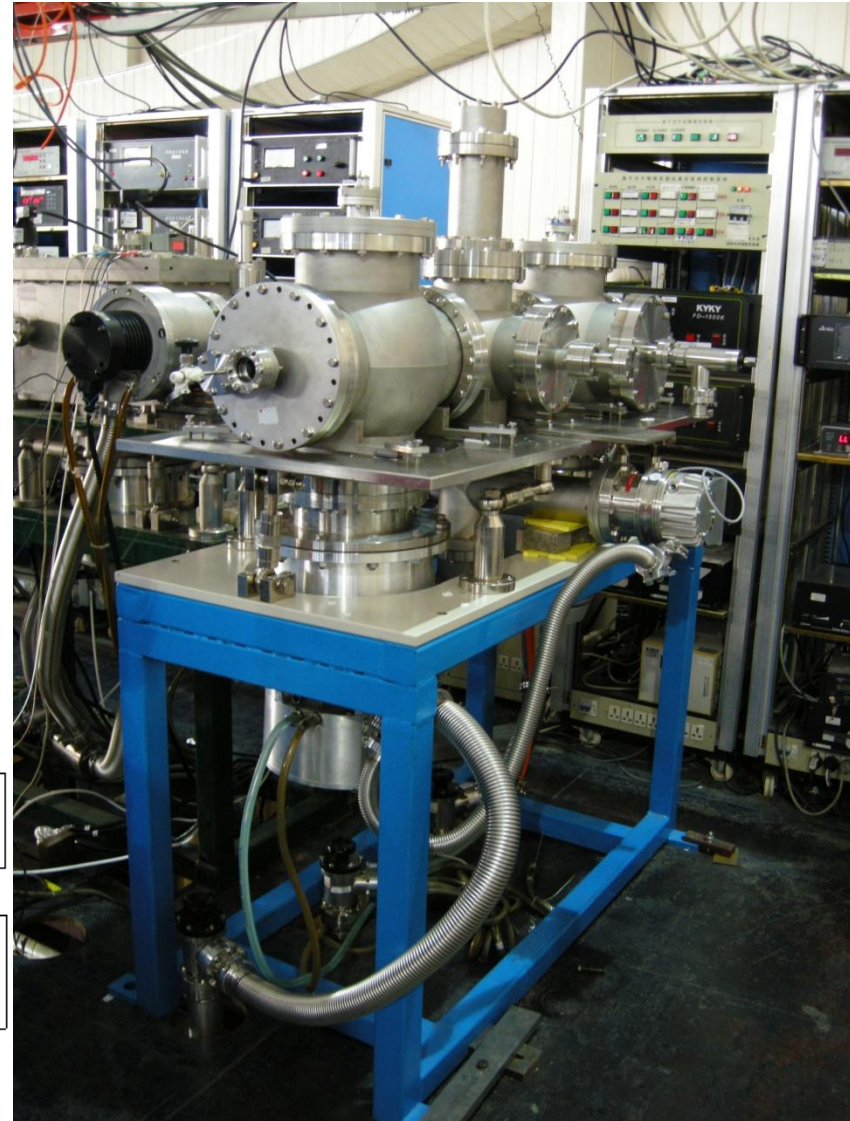
| Ionization region /mm <sup>3</sup> | Hot electron / meV | Ratio of collected hot electrons |               |         |
|------------------------------------|--------------------|----------------------------------|---------------|---------|
|                                    |                    | Ref.1 in 2003                    | Ref.2 in 2002 | Present |
| 1 × 4 × 4                          | 10                 | 26.7%                            | 36.7%         | 3.3%    |
|                                    | 20                 | 18.9%                            | 25.6%         | 2.4%    |
|                                    | 50                 | 12.2%                            | 15.5%         | 1.5%    |



# TPEPICO VMI at Hefei

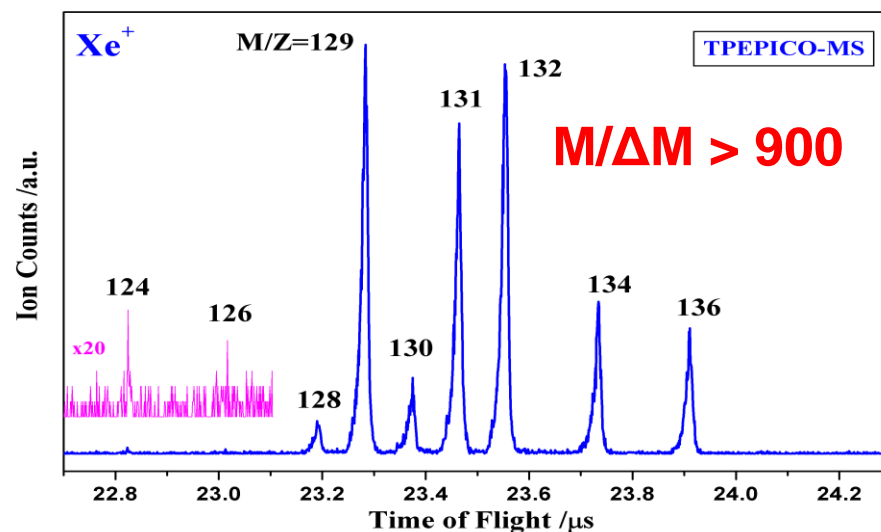
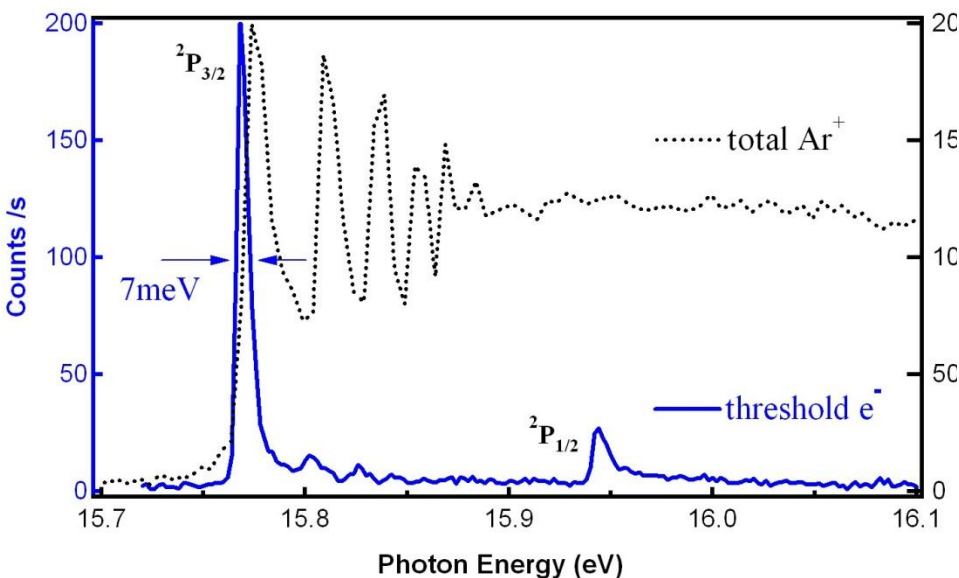
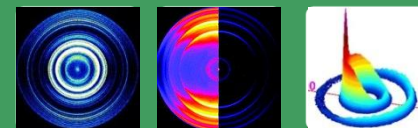


U14A in Hefei Synchrotron Radiation Lab.

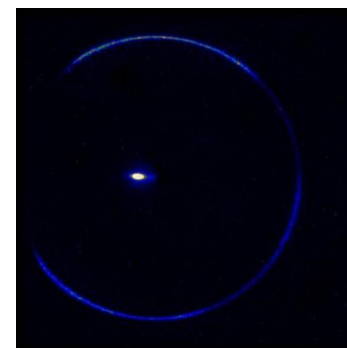


X. F. TANG, X. G. ZHOU\*, ... *Rev. Sci. Instrum.* 80 (2009), 113101

# TPEPICO VMI performance

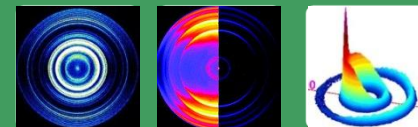


| Performance                              | Hefei                |
|--|----------------------|
| Coincidence counts/s                     | ~ 300                |
| Energy resolution of Threshold $e^-$     | 7 meV<br>Vibrational |
| Velocity map imaging, $\Delta E_t / E_t$ | ~ 2%                 |



$$\frac{\Delta E_T}{E_T} = 2 \times \frac{\Delta v}{v} = 2.2\%$$





## $\text{O}_2^+$ ion:

$\text{B}^2\Sigma_g^-$  vibrational dependence

$^2\Sigma_u^-$  dark ionic state

*J. Phys. Chem. A*, 115, 6339 (2011)

*J. Phys. Chem. A*, 116, 9459 (2012)

## $\text{N}_2\text{O}^+$ ion:

$\text{C}^2\Sigma^+$   $\text{NO}^+$  formation pathways

$\text{N}^+$  formation pathways

*J. Chem. Phys.* 134, 54312 (2011)

in analysis

$\text{D}^2\Pi$  dissociation dynamics

*J. Electron Spectrosc. Relat. Phenom.* (2014)

## $\text{CH}_3\text{Cl}^+$ ion:

$\text{A}^2\text{A}_1$  &  $\text{B}^2\text{E}$  internal conversion

*J. Chem. Phys.* 136, 34304 (2012)

## $\text{CH}_3\text{Br}^+$ ion:

$\text{A}^2\text{A}_1$  &  $\text{B}^2\text{E}$  spin-orbit coupling

*J. Chem. Phys.* 140, 44321 (2014)

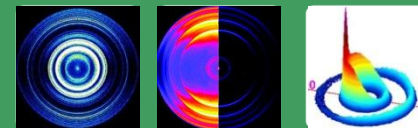
## $\text{CF}_4^+$ ion:

$\text{AP}_0(\text{CF}_3^+/\text{CF}_4) = 14.709 \text{ eV}$

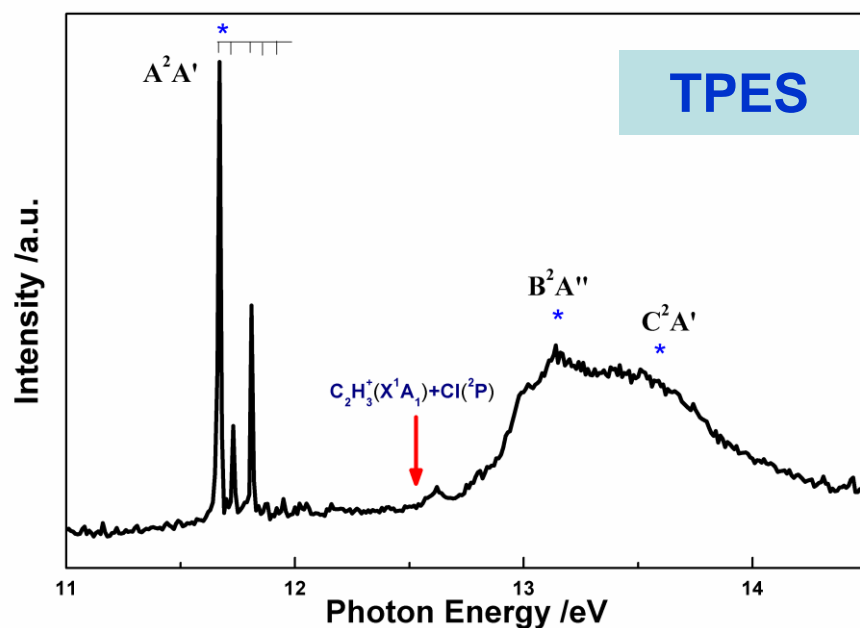
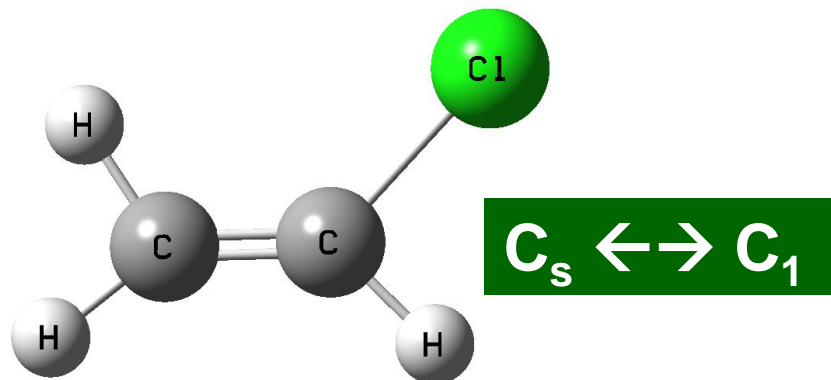
*J. Chem. Phys.* 138, 94306 (2013)

.....

# Cl-loss of $\text{C}_2\text{H}_3\text{Cl}^+$



$$D_e = 12.53 \text{ eV}$$



**$\text{A}^2\text{A}'$  state:**

a series of vibrational bands

**$\text{B}^2\text{A}''$  state:**

from 12.5 eV;

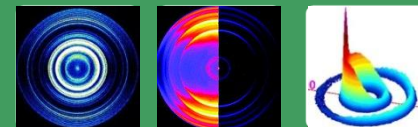
center at 13.14 eV;

no vibrational structure, except  
for the lowest vibronic band

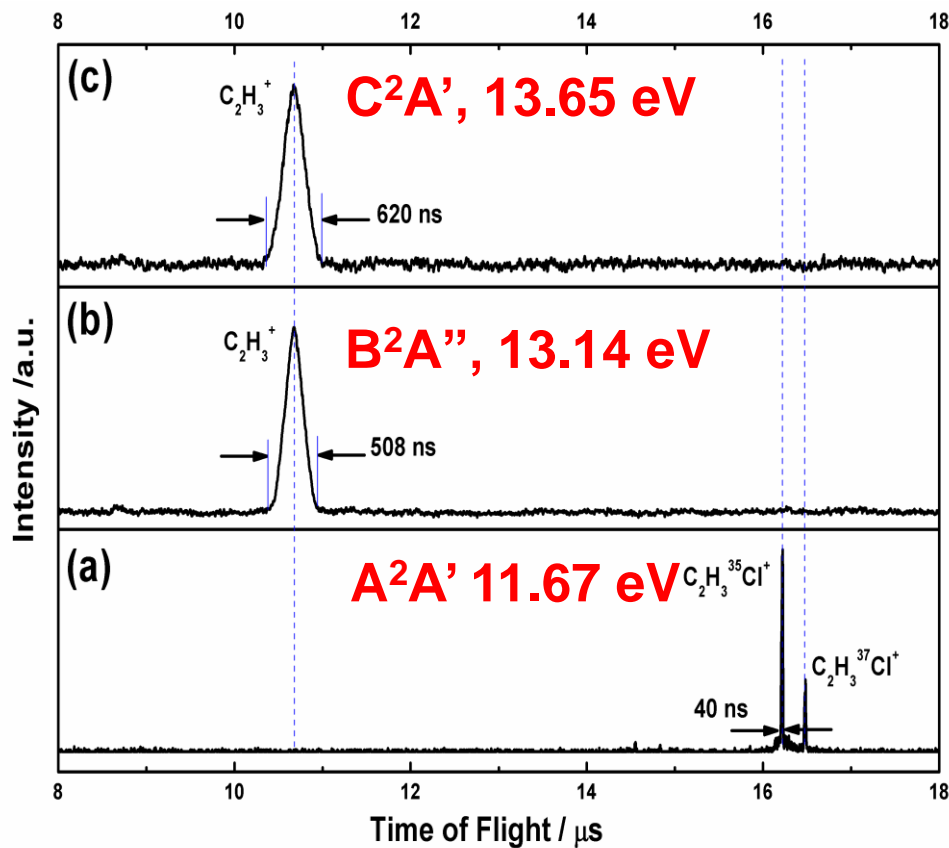
**$\text{C}^2\text{A}'$  state:**

center at  $\sim 13.65$  eV;

no vibrational structure;  
serious overlapping



## TPEPICO TOF Mass Spectra



**$A^2A'$  state: bound**

only  $C_2H_3^{35}Cl^+$  &  $C_2H_3^{37}Cl^+$

**$B^2A''$  state: dissociative**

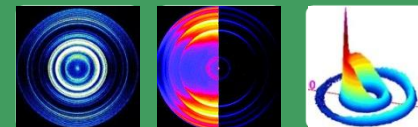
no  $C_2H_3^{35}Cl^+$ ;

$C_2H_3^+$  with 240 ns FWHM

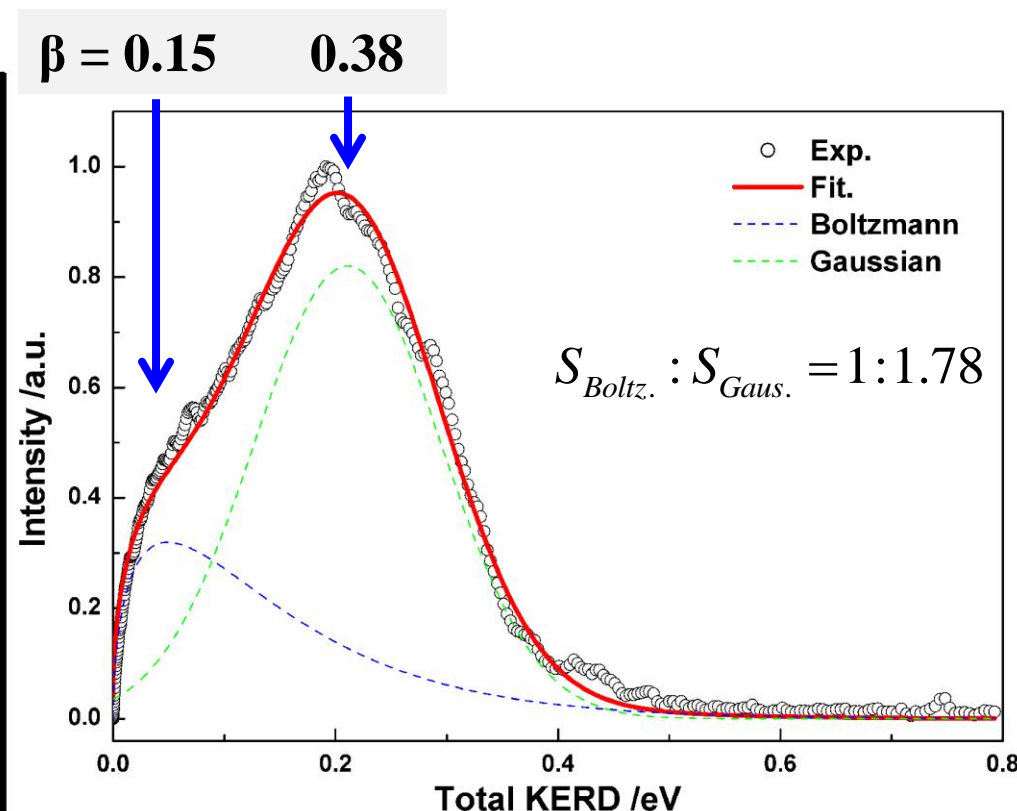
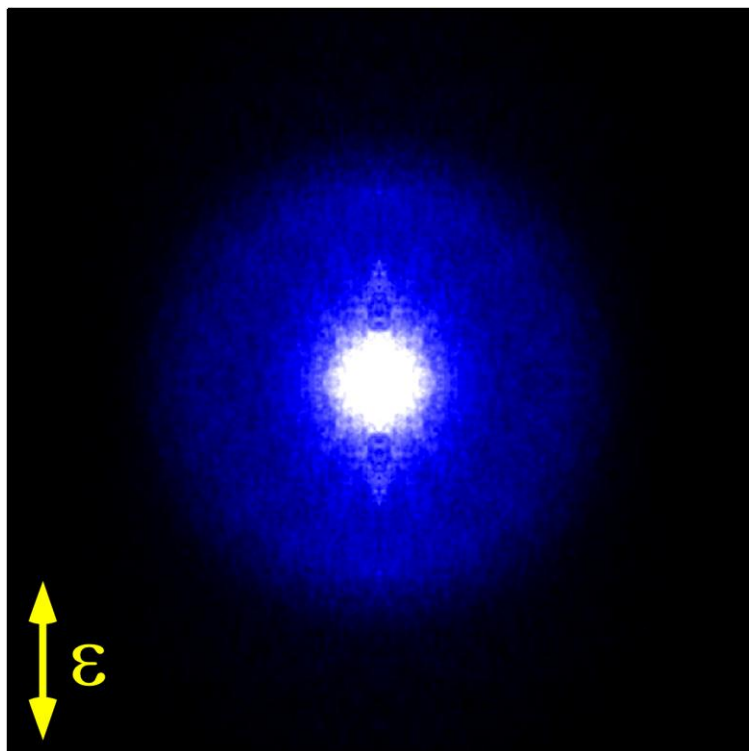
**$C^2A'$  state: dissociative**

no  $C_2H_3^{35}Cl^+$ ;

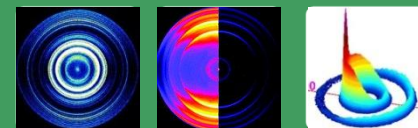
$C_2H_3^+$  with 280 ns FWHM



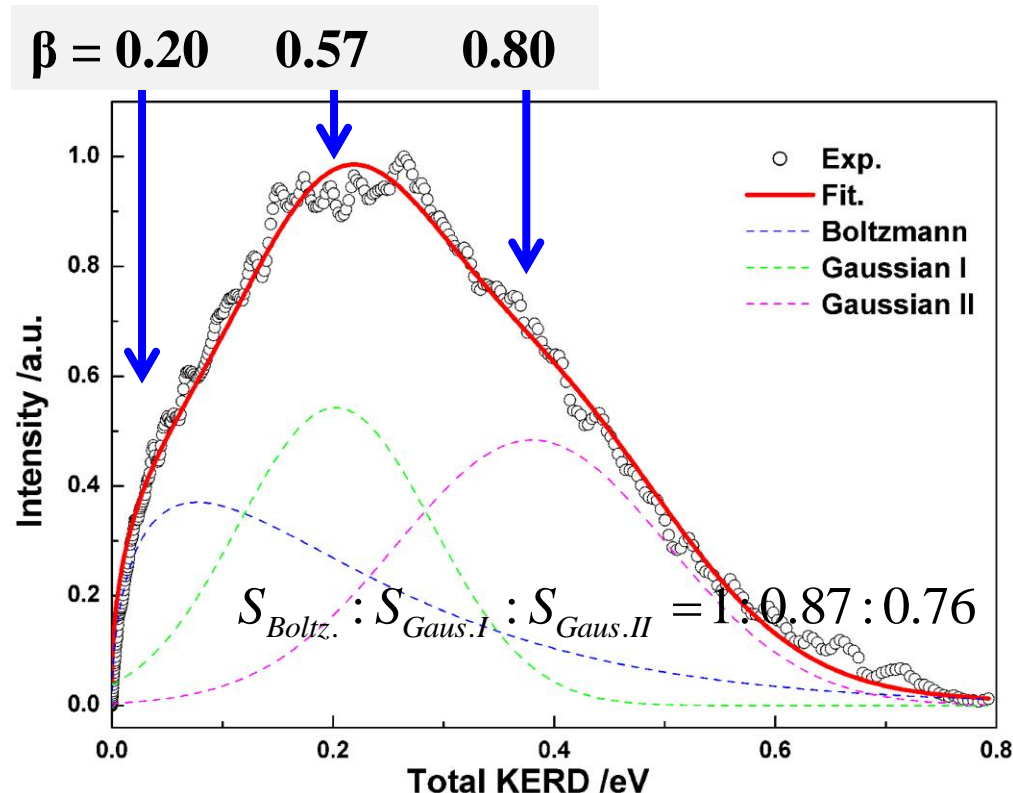
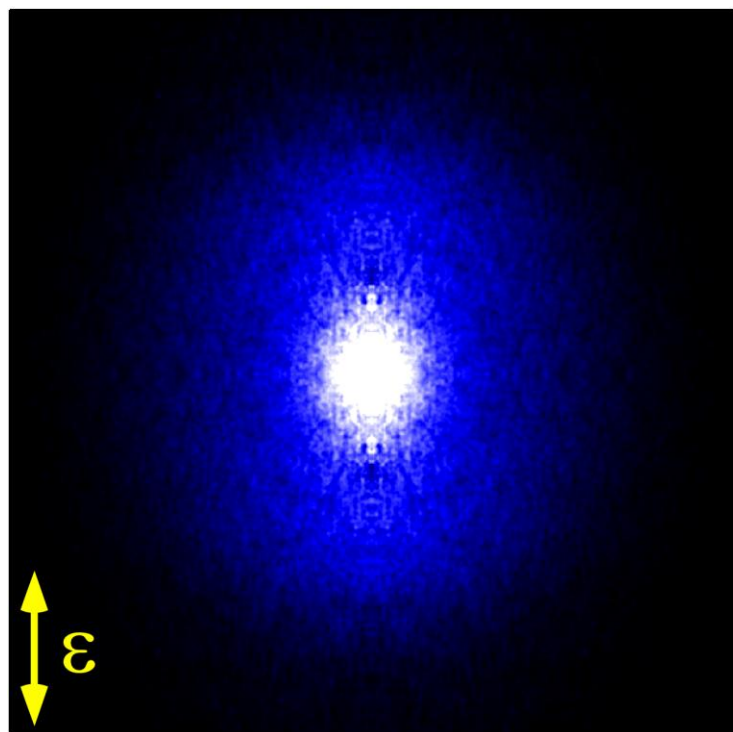
$B^2A''$ , 13.14 eV



- ◆ Bimodal distribution of  $C_2H_3^+$
- ◆ Boltzmann (**statistic**) + Gaussian (**direct dissociation**)

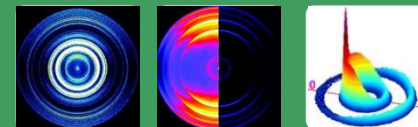


$C^2A'$ , 13.65 eV

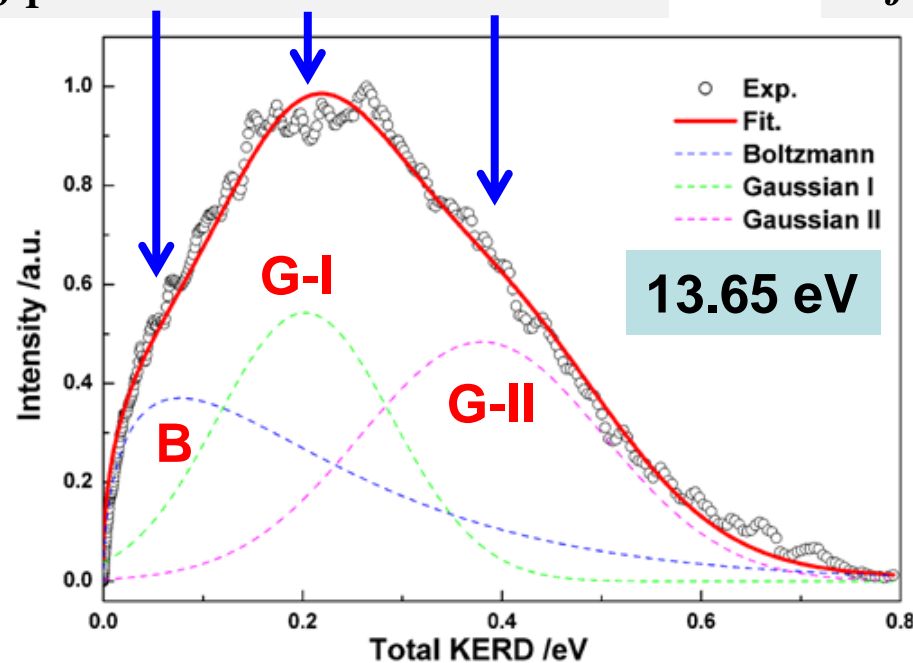


- ◆ Triplet internal energy distribution of  $C_2H_3^+$
- ◆ Boltzmann (**statistic**) + 2 Gaussian (**2 fast dissociation**)
- ◆ A new  $C_2H_3^+$  formation pathway besides those similar to  $B^2A''$

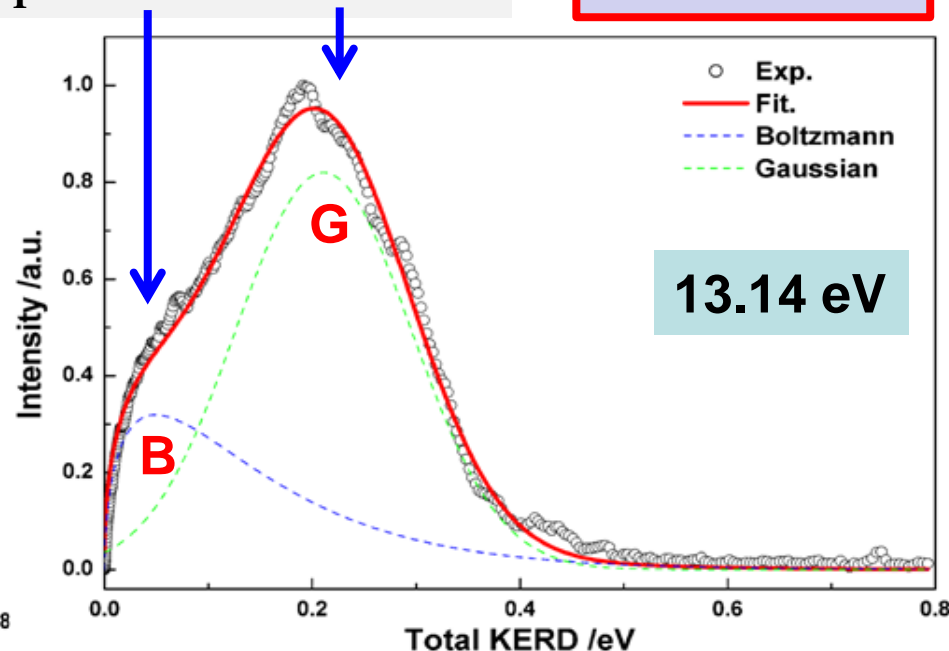
# TKERD curves



$\langle E_T \rangle = 0.15 \quad 0.20 \quad 0.38 \text{ eV}$   
 $f_T = 0.13 \quad 0.18 \quad 0.34$



$\langle E_T \rangle = 0.10 \quad 0.21 \text{ eV}$   
 $f_T = 0.16 \quad 0.34$



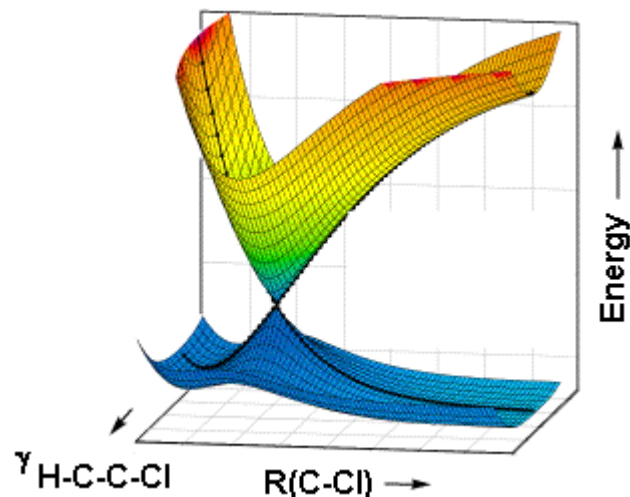
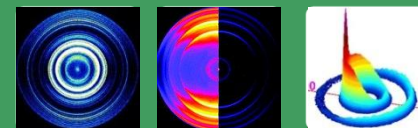
$$f_T = \frac{\langle E_T \rangle}{E_{avail}}$$

At 13.65 eV, **Gauss-II** distribution has the same pathway as at 13.14 eV

What is the Gauss-I population?



# Potential Energy Surface

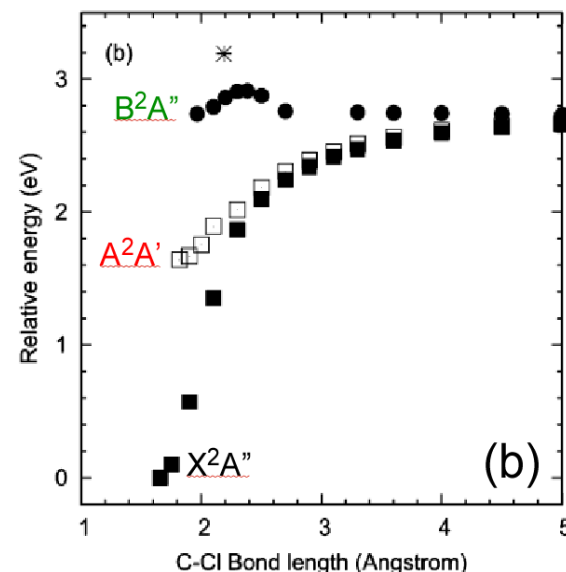
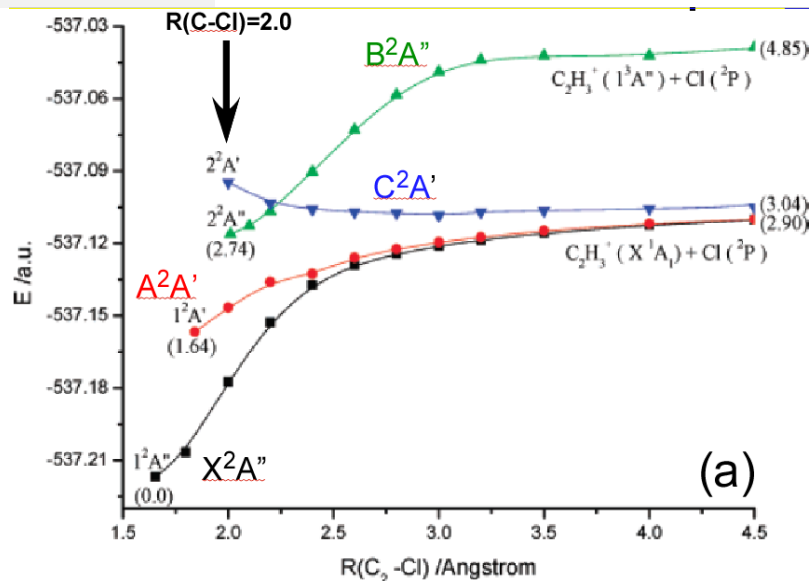


**Conical Intersection  
between B & C states**

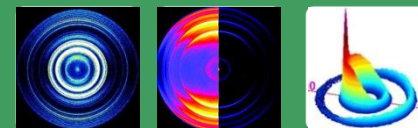
- ◆  **$C_s$  symmetry:**  
 $B^2A''$  bound &  $C^2A'$  repulsive
- ◆ **No symmetry  $C_1$ :**  
 $B^2A''$  predissociative  
 $C^2A'$  unknown

In plane: *J. Phys. Chem. A* 2008, 112, 1688

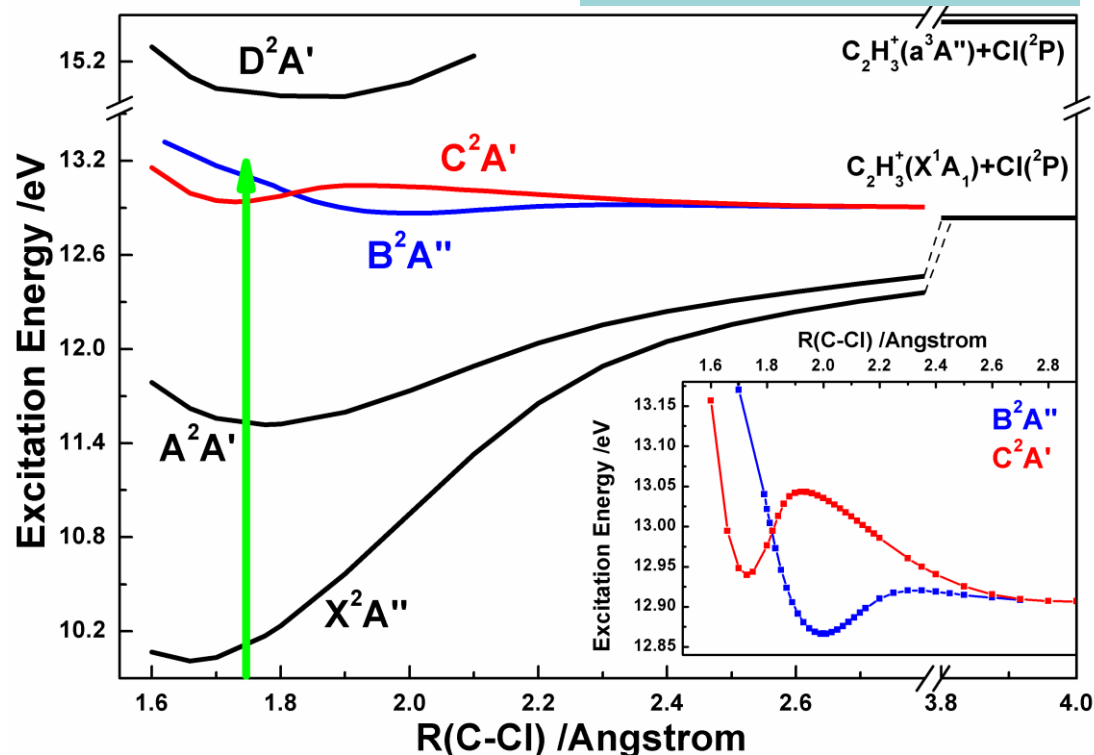
Out of plane: *J. Phys. Chem. A* 2010, 114, 7937



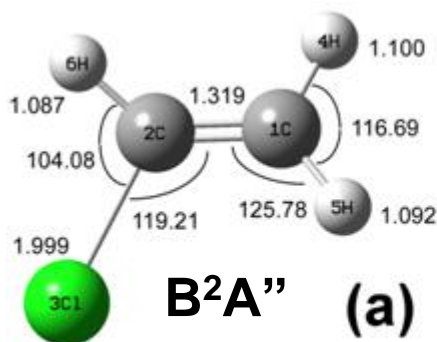
# Potential Energy Surface



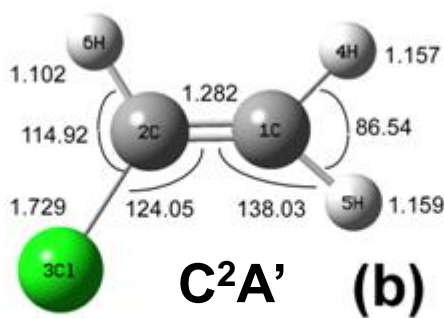
at TD-B3LYP/6-311+G(d,p)



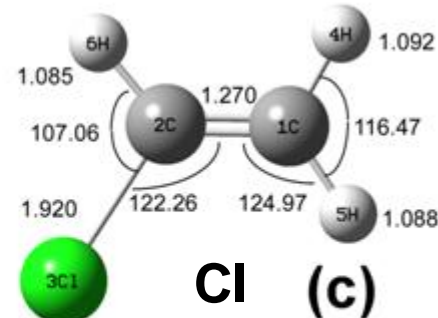
- ◆ No symmetry  $C_1$ :
- $B^2A''$  &  $C^2A'$  predissociative
- ◆ Geometries:
- $B^2A''$  bent
- $C^2A'$  planar
- ◆ Cl between B & C:
- 0.13 eV above  $B^2A''$



$$D(^3Cl-^2C-^1C-^4H)=161.2$$

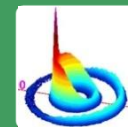
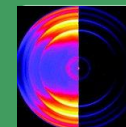
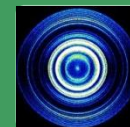


$$D(^3Cl-^2C-^1C-^4H)=180.0$$



$$D(^3Cl-^2C-^1C-^4H)=180.0$$

# Potential Energy Surface



| State                               | R(C-C) /Å          | R(C-Cl) /Å         | A( <sup>1</sup> C- <sup>2</sup> C- <sup>6</sup> H) <sup>a</sup> | T <sub>v</sub> /eV <sup>b</sup> |                     | T <sub>0</sub> /eV <sup>b</sup> |                      |
|-------------------------------------|--------------------|--------------------|---|---------------------------------|---------------------|---------------------------------|----------------------|
|                                     |                    |                    |   | Cal.                            | Exp.                | Cal.                            | Exp.                 |
| X <sup>2</sup> A''(C <sub>s</sub> ) | 1.395              | 1.659              | 122.54  | 10.10                           | 10.005 <sup>d</sup> | 9.81                            | 10.0062 <sup>e</sup> |
|                                     | 1.405 <sup>c</sup> | 1.657 <sup>c</sup> | 122.5 <sup>c</sup>  | 9.631 <sup>c</sup>              |                     |                                 |                      |
|                                     | 1.392 <sup>e</sup> | 1.657 <sup>e</sup> | 122.4 <sup>e</sup>  |                                 |                     |                                 |                      |
|                                     | 1.391 <sup>f</sup> | 1.640 <sup>f</sup> | 122.6 <sup>f</sup>  |                                 |                     |                                 |                      |
| A <sup>2</sup> A'(C <sub>s</sub> )  | 1.326              | 1.775              | 128.41  | 11.75                           | 11.664 <sup>d</sup> | 11.74                           | 11.6667 <sup>e</sup> |
|                                     | 1.327 <sup>f</sup> | 1.752 <sup>f</sup> | 131.2 <sup>f</sup>  | 11.141 <sup>c</sup>             |                     |                                 |                      |
| B <sup>2</sup> A''(C <sub>1</sub> ) | 1.319              | 1.999              | 135.74  | 13.07                           | 13.13 <sup>d</sup>  | 12.84                           | 12.7518 <sup>e</sup> |
|                                     | 1.368 <sup>c</sup> | 1.964 <sup>c</sup> | 131.6 <sup>c</sup>  | 12.596 <sup>c</sup>             |                     |                                 |                      |
|                                     | 1.316 <sup>e</sup> | 1.991 <sup>e</sup> | 135.6 <sup>e</sup>  |                                 |                     |                                 |                      |
|                                     | 1.354 <sup>f</sup> | 1.937 <sup>f</sup> | 131.4 <sup>f</sup>  |                                 |                     |                                 |                      |
| C <sup>2</sup> A'(C <sub>s</sub> )  | 1.282              | 1.729              | 121.03  | 13.57                           | 13.56 <sup>d</sup>  | 13.17                           | --                   |
|                                     |                    |                    |   | 13.422 <sup>c</sup>             |                     |                                 |                      |
| X <sup>1</sup> A'(C <sub>s</sub> )  | 1.324              | 1.752              | 124.40  | --                              | --                  | --                              | --                   |
|                                     | 1.332 <sup>g</sup> | 1.726 <sup>g</sup> | 123.8 <sup>g</sup>  |                                 |                     |                                 |                      |

a. C=C-H bond angle (degrees) in vinyl cation moiety.

b. ionization energies were calculated with the optimized geometry of ground neutral molecule.

c. SA-CASSCF optimized geometries and MRCISD energies from Ref. 21.

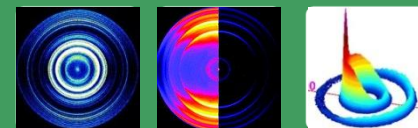
d. from He(I) photoelectron spectra of Ref. 9.

e. from MATI-PFY spectra of Ref. 12.

f. CASPT2 geometries from Ref. 20.

g. experimental data from Ref. 42.

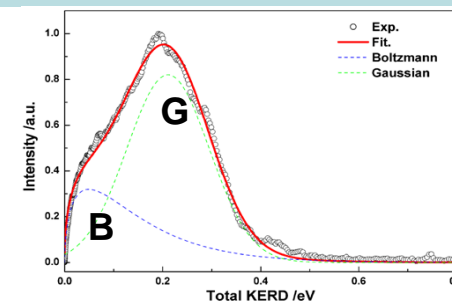
# Dissociative Photoionization Mechanism



◆ at 13.14 eV:  $B^2A''$  no contribution from  $C^2A'$

◆ Boltzmann: IC to lower state + statistical, slow

◆ Gauss: direct dissociate along  $B^2A''$ , fast

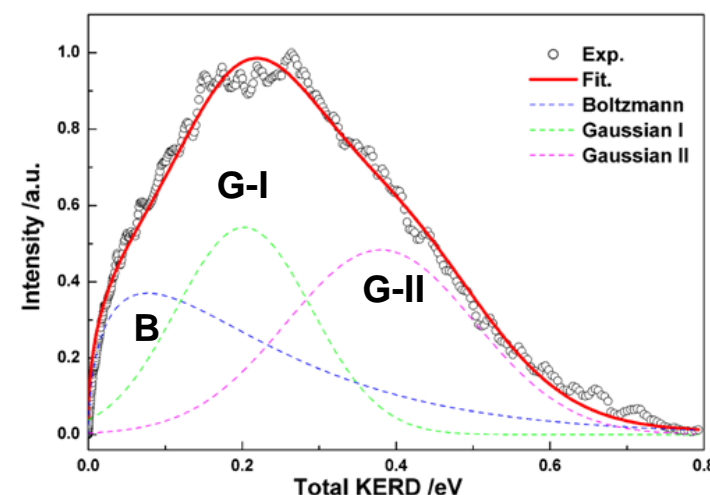


◆ at 13.65 eV:  $B^2A''$  with high  $v^+$  &  $C^2A'$  due to overlap

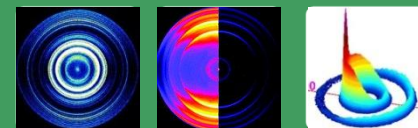
◆ Boltz. & Gauss-II: similar to 13.14eV

◆ Gauss-I: coupling from  $C^2A'$  to  $B^2A''$  by conical intersection, slow

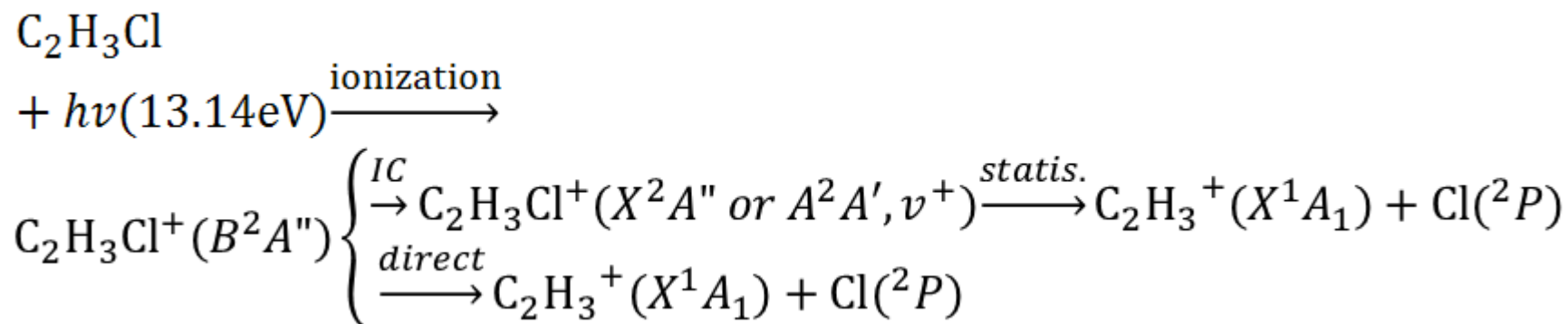
IVR in the following IC process to cause some excess energy redistributed to ro-vibrational excited  $C_2H_3^+$



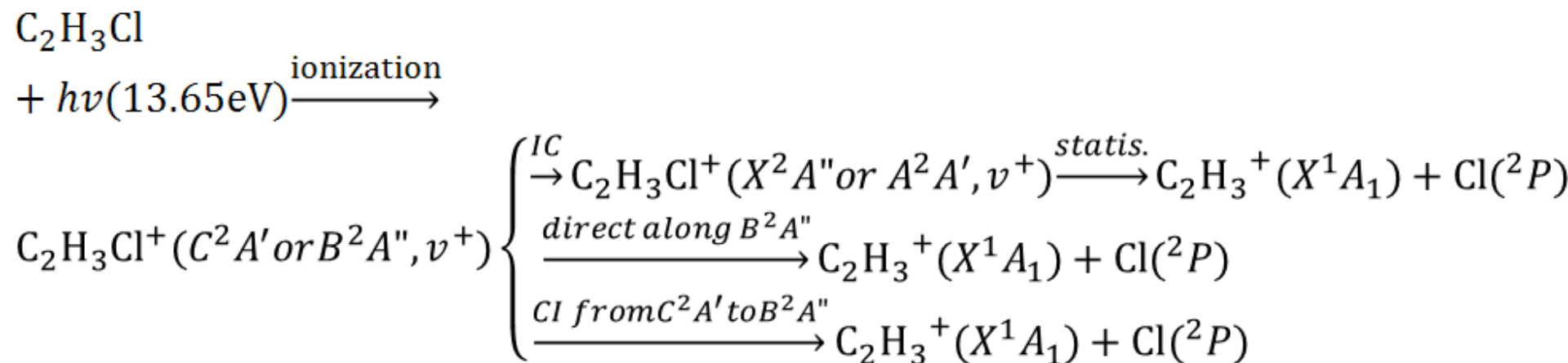
# Conclusions



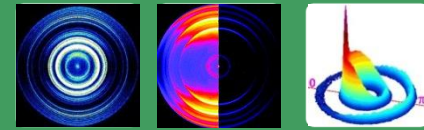
## ◆ at 13.14 eV: $B^2A''$



## ◆ at 13.65 eV: $B^2A''$ with high $v^+$ & $C^2A'$



# Acknowledgments



## ◆ Co-workers

◆ *Prof. Shilin Liu, Dr. Xiaofeng Tang*

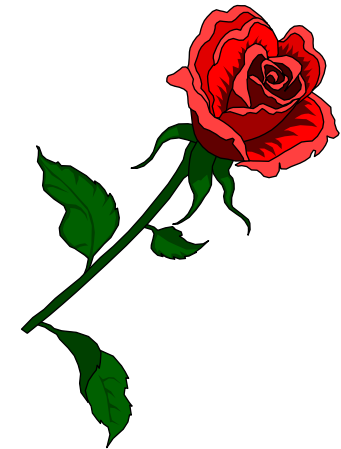
◆ *Mr. Mingli Niu, Ms. Manman Wu*

## ◆ Collaborators

◆ *Prof. Liusi Sheng, Dr. Fuyi Liu, Mr. Xiaobin Shan*

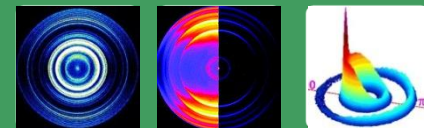
(National Synchrotron Radiation Laboratory at Hefei)

## ◆ Funds

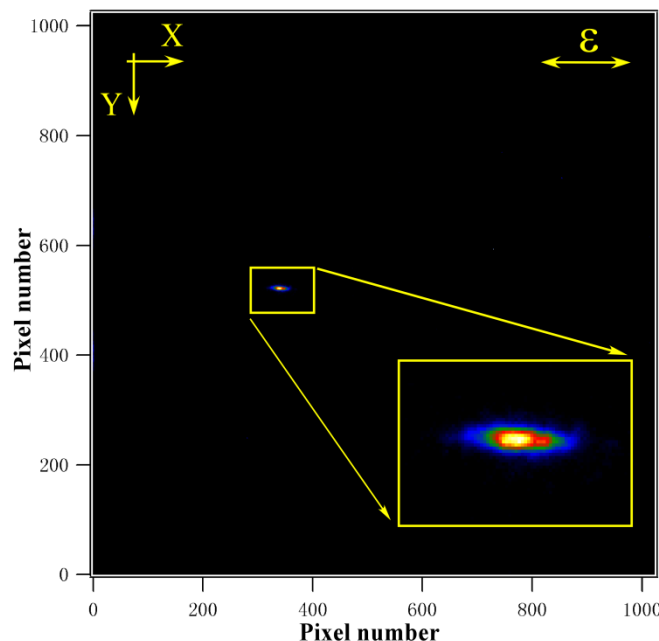




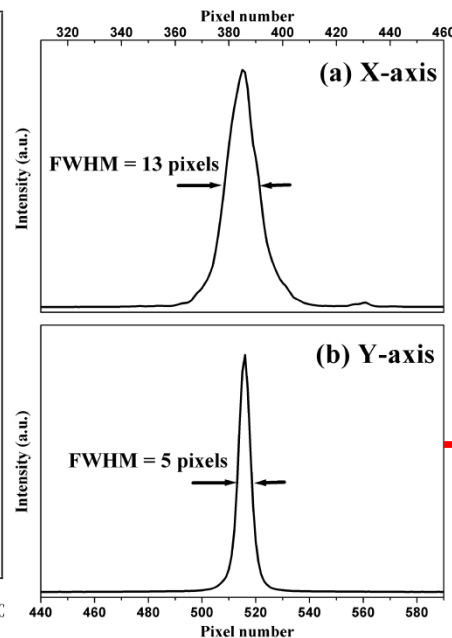
# Deconvolution of image



$\text{Ar}^+$

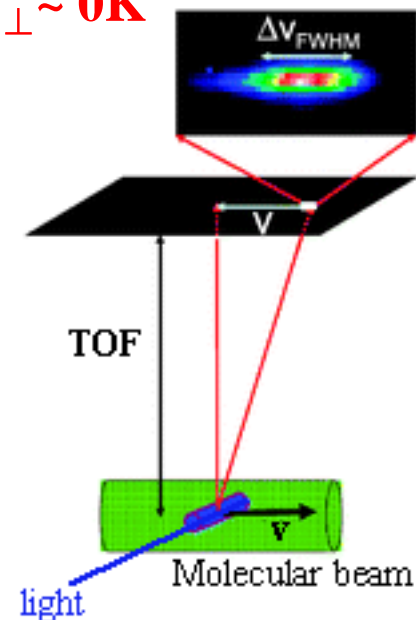


Molecular beam direction

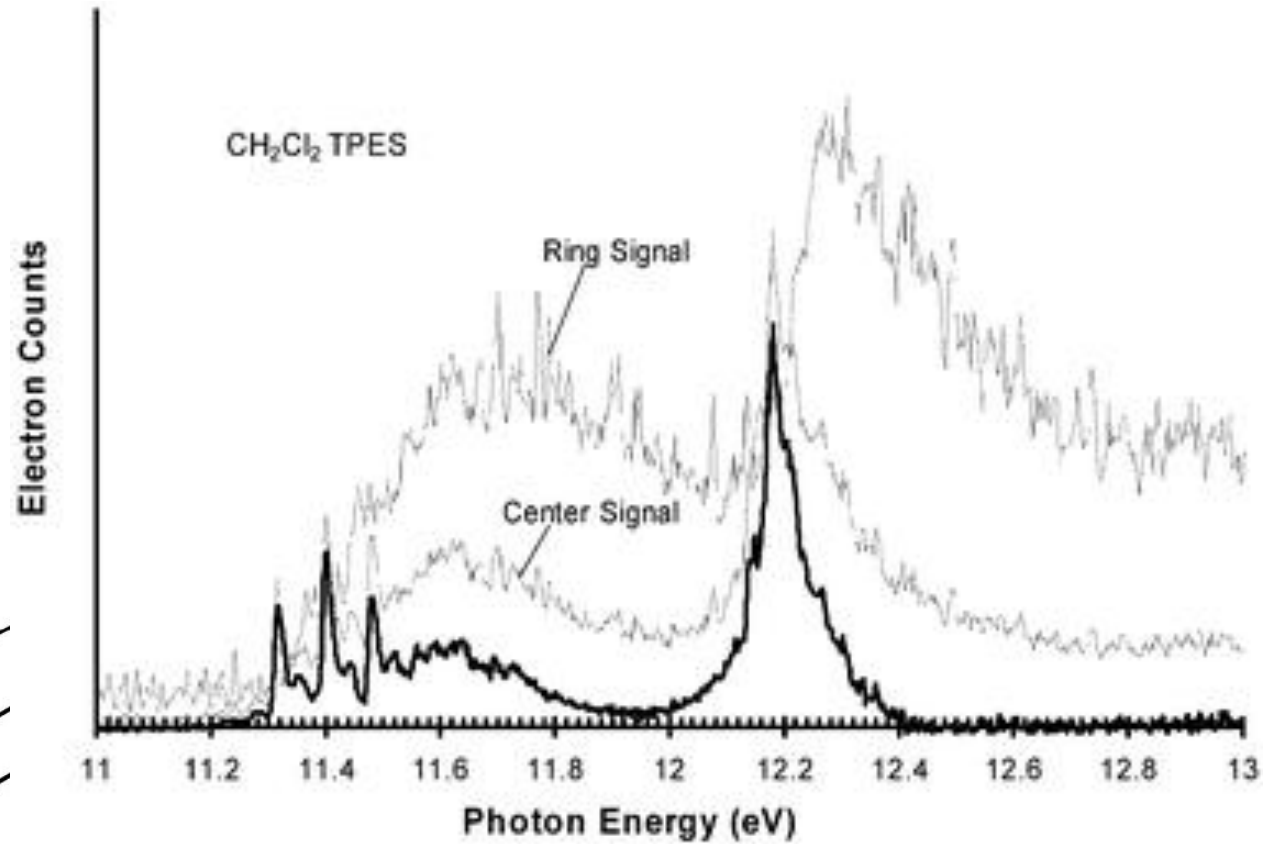
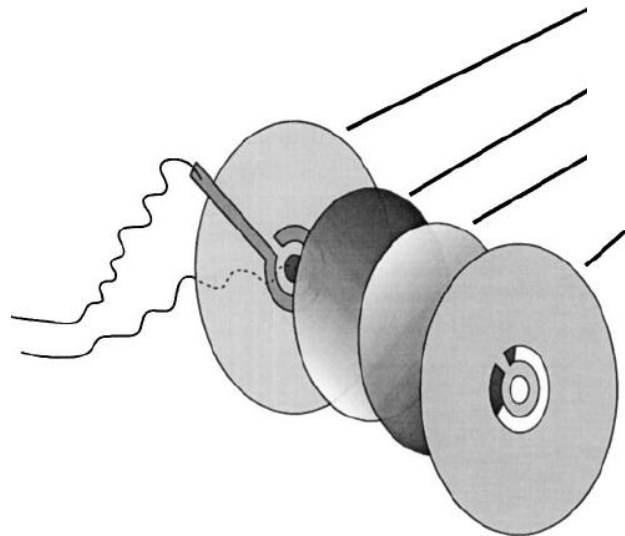
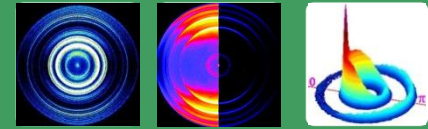


$\rightarrow T_{//} \sim 7\text{K}$

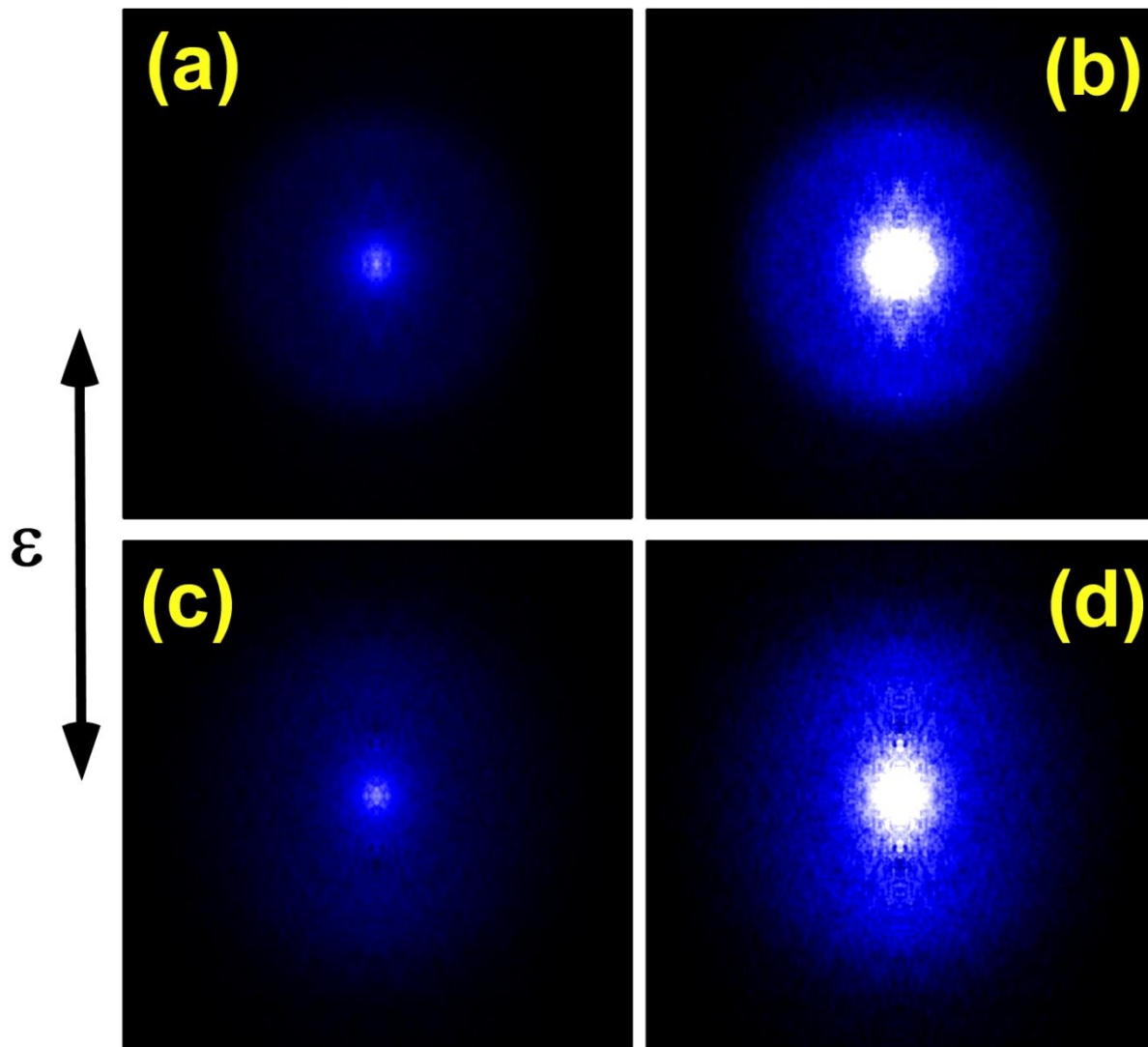
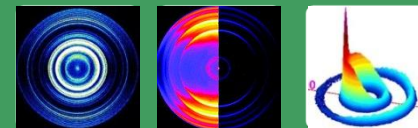
$\rightarrow T_{\perp} \sim 0\text{K}$



# Subtraction mode



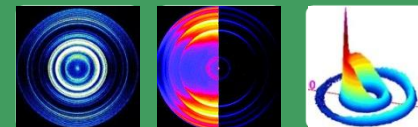
T. Baer *et al.*, *Rev. Sci. Instrum.*, 74, 3763(2003)



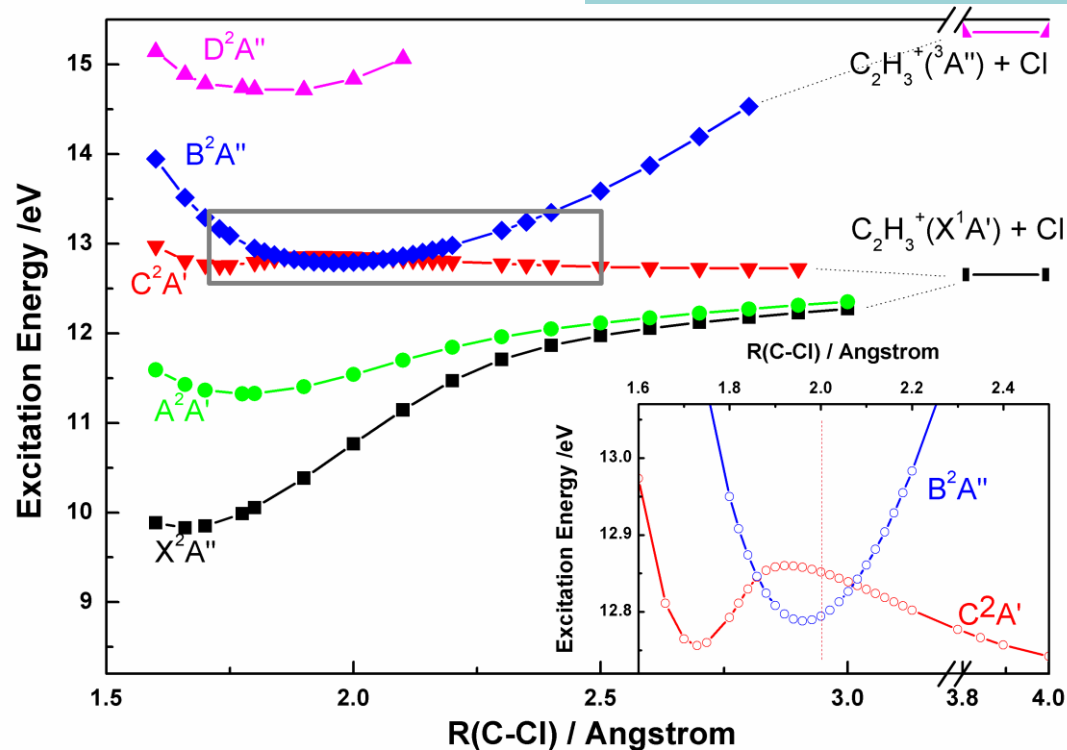
**B<sup>2</sup>A'', 13.14 eV**

**C<sup>2</sup>A', 13.65 eV**

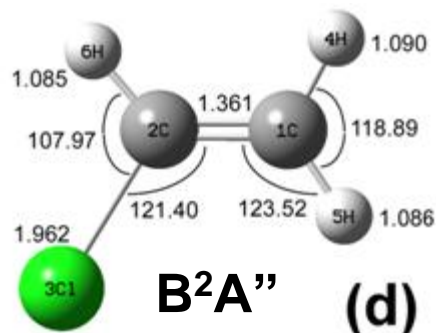
# Potential Energy Surface



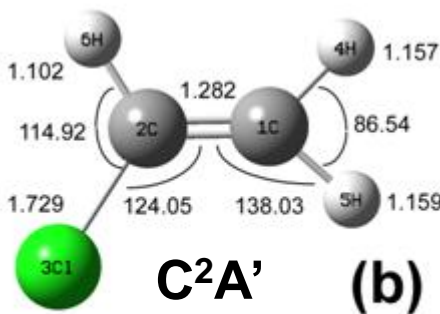
at TD-B3LYP/6-311+G(d,p)



- ◆  **$\text{C}_s$  symmetry:**
  - $\text{B}^2\text{A}''$  bound
  - $\text{C}^2\text{A}'$  predissociative
- ◆ **Geometries:**
  - $\text{B}^2\text{A}''$  &  $\text{C}^2\text{A}'$  planar



$\text{D}(\text{}^3\text{Cl}-\text{}^2\text{C}-\text{}^1\text{C}-\text{}^4\text{H}) = 180.0$



$\text{D}(\text{}^3\text{Cl}-\text{}^2\text{C}-\text{}^1\text{C}-\text{}^4\text{H}) = 180.0$