

THE ELECTRONIC STRUCTURE OF THE PLANARIZED BLATTER RADICAL AND ITS DERIVATIVES.

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In recent years, derivatives of the 1,4-dihydrobenzo[e][1,2,4]triazin-4-yl radical (e.g. the Blatter1 radical^a) are gaining much interest due to their properties, such as exceptional stability, spin Π -delocalization, narrow electrochemical window and low excitation energies. For these reasons, a rapidly increasing attention is given to these radicals as structural elements of advanced materials.^b Their design requires, however, a good understanding of the electronic structure of these electro-, photo- and magnetically active molecular components.

Recent advances in the chemistry of the 1,4-dihydrobenzo[e][1,2,4]triazin-4-yl demonstrated access to the parent “planarized” Blatter radical, in which more effective spin delocalization onto the Ph ring at the N(1) position is observed.^c In order to better understand the impact of electronic delocalization on properties of the radicals, a series of derivatives has been prepared and investigated by computational and spectroscopy methods.

Herein we present determination of the electronic structure of a series of substituted planar radical using UV photoelectron spectroscopy (UV-PES), EPR, and UV-vis spectroscopy.

^aBlatter, H. M.; Lukaszewski, H. A New Stable Free Radical. *Tetrahedron Lett.* 1968, 9, 2701–2705.

^bJasiński, M.; Szczytko, J.; Pocięcha, D.; Monobe, H.; Kaszyński, P. “Substituent-dependent magnetic behavior of discotic benzo[e][1,2,4]triazinyls”, *J. Am. Chem. Soc.* 2016, 138, 9421–9424. Kapuściński, S.; Gardias, A.; Pocięcha, D.; Jasiński, M.; Szczytko, J.; Kaszyński, P. “Paramagnetic bent-core mesogens derived from the 1,4-dihydrobenzo[e][1,2,4]triazin-4-yl”, *J. Mater. Chem. C*, 2018, 6, 3079–3088.

^cKaszyński, P.; Constantinides, C. P.; Young, V. G. The Planar Blatter Radical: Structural Chemistry of 1,4-Dihydrobenzo[e][1,2,4]triazin-4-yls. *Angew. Chem.* 2016, 128, 11315–11318.