

SPECTROSCOPIC STUDIES OF SMALL POLYATOMIC MOLECULES NEAR ION-PAIR DISSOCIATION THRESHOLDS

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At high vibrational quantum numbers, ion-pair states (A^+-B^-) behave like high Rydberg states. Their level structure can be approximately described by Rydberg's formula after replacing the Rydberg constant R_∞ by the appropriate mass-scaled value $R_{IP} = R_\infty \frac{\mu_{A^+-B^-}}{m_e}$.^a

We present the results of spectroscopic investigations of high-lying excited ion-pair states of small polyatomic molecules in the vicinity of ion-pair dissociation thresholds. Such states have been extensively characterized in diatomic molecules (see ref. ^{a,b} and references therein) and are at the heart of the technique of threshold ion-pair production spectroscopy.^c Only few studies have been carried out on polyatomic molecules.^{d,e} Polyatomic molecules exhibit denser series of ion-pair dissociation thresholds corresponding to the different rovibrational levels of the positively and negatively charged fragments. We will present the results of our investigations of OCS and H₂S in the vicinity of their lowest ion-pair dissociation thresholds. The experiments are carried out using cold samples in skimmed supersonic expansions. The region of the lowest ion-pair dissociation threshold is reached by single-photon excitation from the neutral ground state using narrow-band VUV radiation. The experiments provide access to the ion-pair dissociation energies from which the level structure of the ionic fragments can be determined as well as other thermochemical data such as electron affinities, ionization and dissociation energies.

^aE. Reinhold and W. Ubachs, *Mol. Phys.*, 103:10, 1329-1352, (2005)

^bS. Mollet, F. Merkt, *Phys. Rev. A*, 82, 032510, (2010)

^cJ.D.D. Martin, J.W. Hepburn, *Phys. Rev. Lett.*, 79, 3154, (1997)

^dR. C. Shiell, X. K. Hu, Q. J. Hu, J. W. Hepburn, *J. Phys. Chem.*, 104, 19 (2000)

^eQ. J. Hu, Q. Zhang, J.W. Hepburn, *J. Chem. Phys.*, 123, 074310, (2006)