



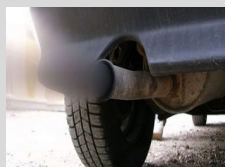
# **Vibrational spectroscopy and reactivity of ultra-small silica and silicate fragments in the gas-phase**

Sandra M. Lang

Institute of Surface Chemistry and Catalysis, Ulm University, Germany



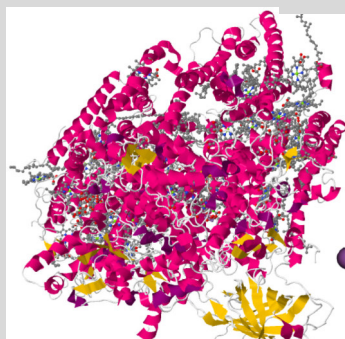
# Modeling the Active Centers of Catalysts by Small Clusters in the Gas Phase



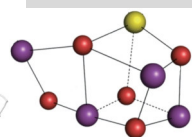
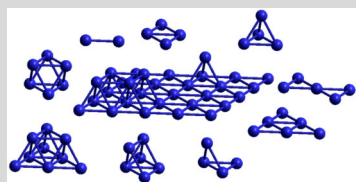
carazoo.com



seilnacht.com



PDB entry 1S5L



$\text{CaMn}_4\text{O}_5$

## Biocatalysts

### CaMnO

Water Oxidation



### FeS

Nitrogen Fixation



## Heterogeneous catalysts

### Copper

$\text{CO}_2$  Conversion



### Gold

Hydrocarbon  
Activation



### Palladium

CO Oxidation



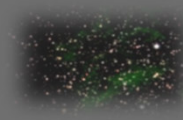
### Ruthenium

CO Methanation

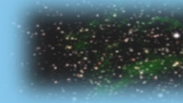


## Astrochemically relevant materials

### MgSiO



### FeS





## Methods

### Experiments

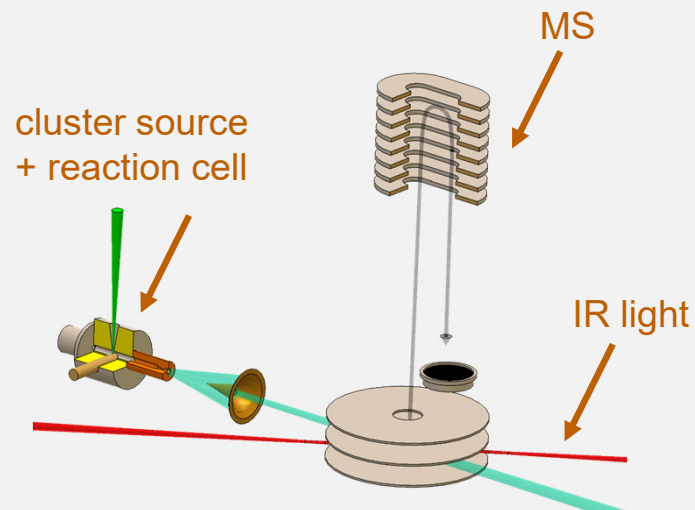
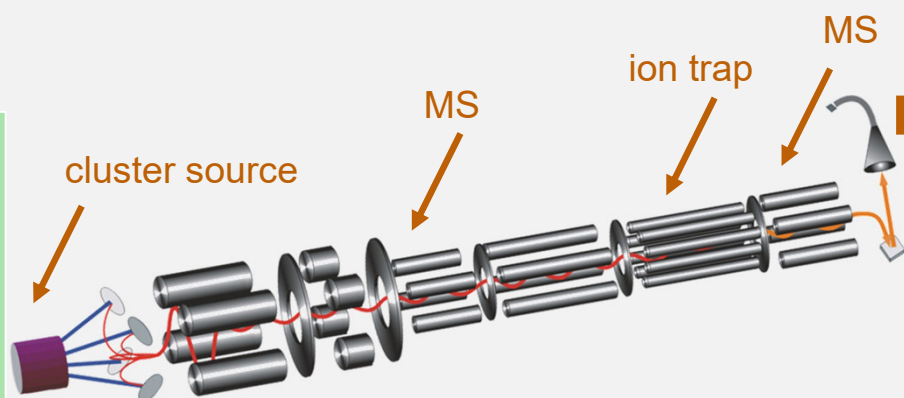
#### Ion Trap Reactivity (Ulm)



#### IR-MPD Spectroscopy (Nijmegen)



collaboration with J. Bakker



### Theory

#### Theory



collaboration with  
S. Bromley  
U. Landman  
T. Hözl



## Free Electron Lasers @ FELIX Laboratory Nijmegen

**FELIX** - Free Electron Laser for  
Infrared EXperiments

**FELICE** - Free Electron Laser for  
IntraCavity EXperiments

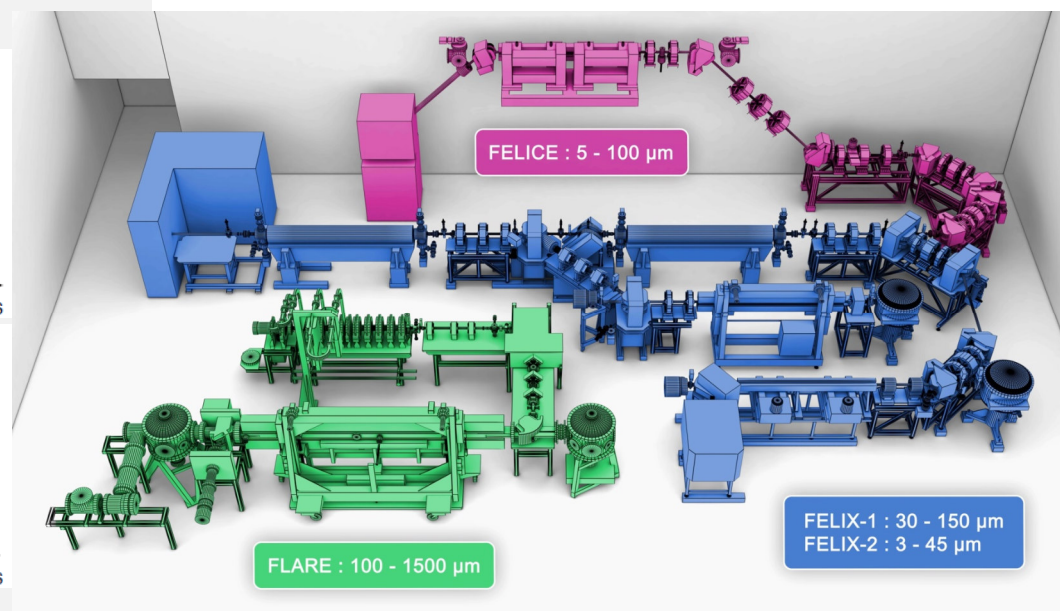
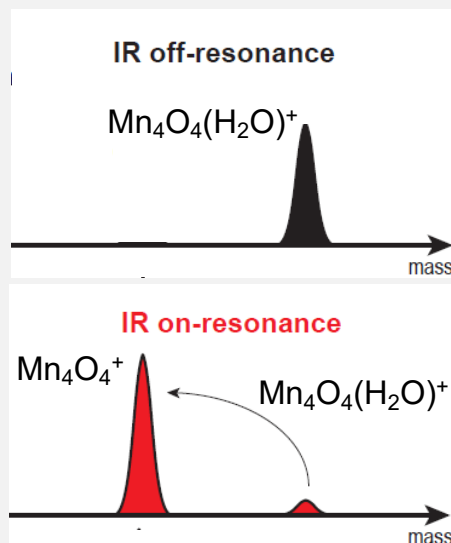
spectral range: 5-100  $\mu\text{m}$   
2000-100  $\text{cm}^{-1}$

- molecule activation
- metal-metal
- metal-oxygen

pulse energy: 5J @ 1 GHz

- strongly bound species

Infrared **M**ulti-**P**hoton **D**issociation  
= action spectroscopy





## Production of Magnesium Silicates

### silicate dust:

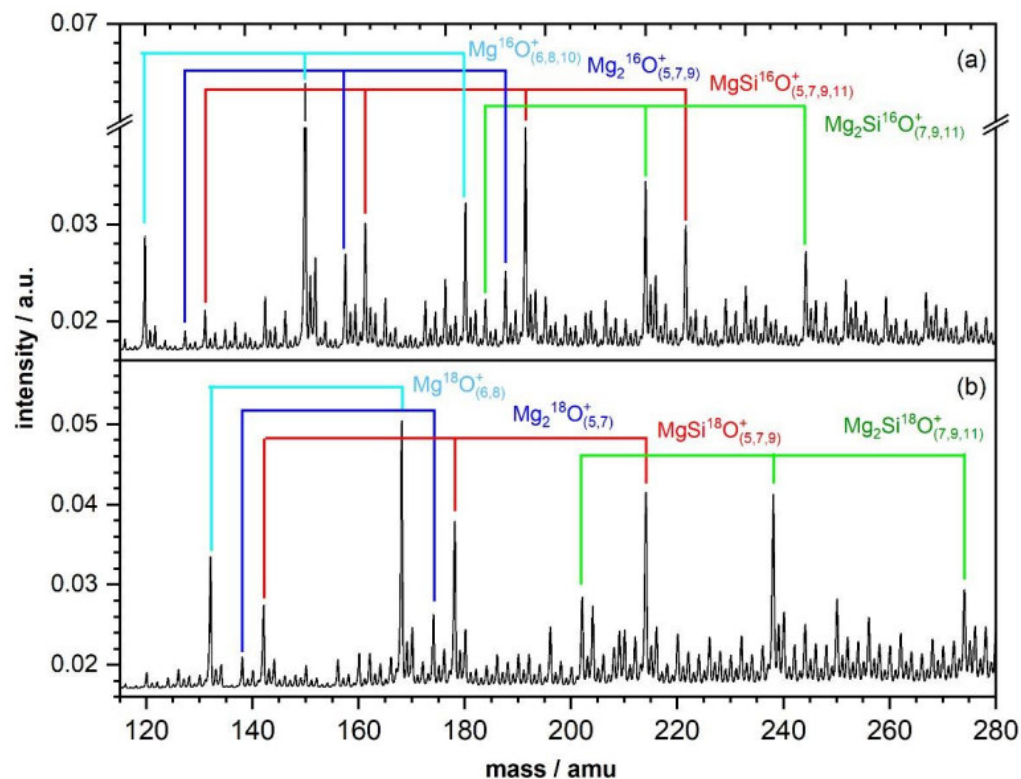
- pyroxene (enstatite) composition ( $\text{MgSiO}_3$ )
- olivine (forsterite) composition ( $\text{Mg}_2\text{SiO}_4$ )

### identification of

- $\text{MgO}_{6,8,10}^+ / \text{Mg}_2\text{O}_{5,7,9}^+$
- $(\text{MgSiO}_3)(\text{O}_2)_{1,2,3,4}^+ \rightarrow$  pyroxene??
- $(\text{Mg}_2\text{SiO}_3)(\text{O}_2)_{2,3,4}^+ \rightarrow$  structure??

→ highly oxidized clusters

laser ablation of a Mg:Si = 2:1 target @ 1%  $\text{O}_2/\text{He}$





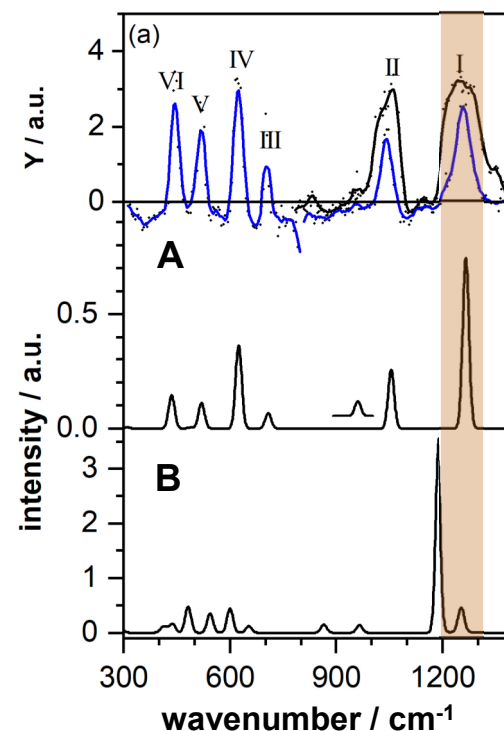
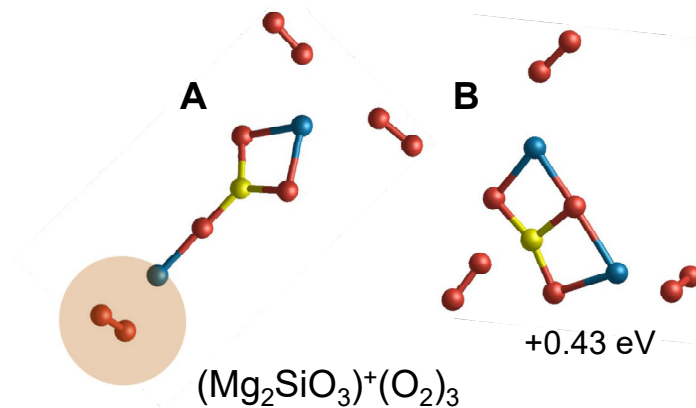
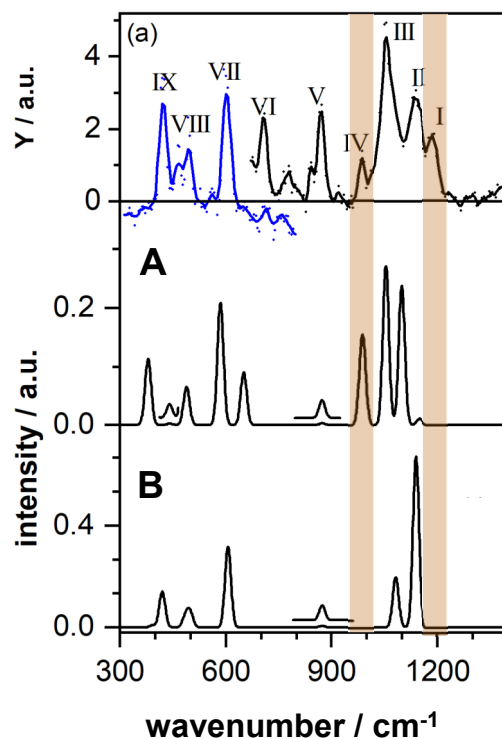
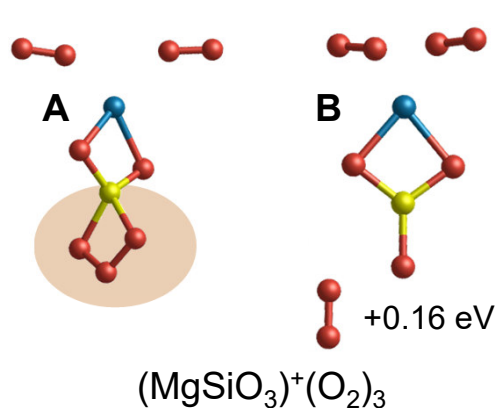
## Structure of Pyroxene (MgSiO<sub>3</sub>)<sup>+</sup> Monomers

- (MgSiO<sub>3</sub>)<sup>+</sup> are structurally robust
- cationic (MgSiO<sub>3</sub>)<sup>+</sup> readily adsorb O<sub>2</sub>
- rather strong binding of oxygen
  - 0.48–0.6 eV
  - (MgSiO<sub>3</sub>)<sup>+</sup>(O<sub>2</sub>)<sub>3</sub>: ozone like O<sub>3</sub>: 4.3 eV
  - (Mg<sub>2</sub>SiO<sub>3</sub>)<sup>+</sup>(O<sub>2</sub>)<sub>3</sub>: super-oxo like O<sub>2</sub>: 1.47 eV
- amorphous silicates: ~0.08 eV

→ **new oxygen reservoir??**

- Mg atom binds to (MgSiO<sub>3</sub>)<sup>+</sup> core

→ **simple initial step in grain growth??**





## Summary

### Combining Reaction Studies and IR Spectroscopy

- determination of cluster structures
- insight in cluster growth process
- insight into cluster chemistry

Ion Trap  
Reactivity

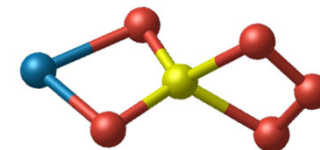


IR-MPD  
Spectroscopy



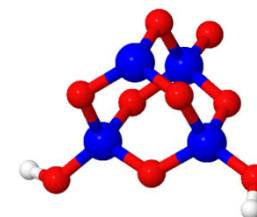
### Pyroxene ( $\text{MgSiO}_3$ )<sup>+</sup> Monomers

- gas-phase formation of pyroxene ( $\text{MgSiO}_3$ )<sup>+</sup> monomers
- facile and strong adsorption of oxygen (ozone- and super-oxo-like)
- binding of Mg atom



### Ultra-Small Silica Clusters

- dissociation of water
- stabilization of relaxed 3D structure through hydroxylation
- characteristic band at 1200-1150  $\text{cm}^{-1}$ ?
- dissociation of  $\text{CO}_2$





## Acknowledgements

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Prof. Tibor Hölzl, FETI, Hungary  
Prof. László Nyulászi, Budapest, Hungary

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Prof. Dr. Joost M. Bakker  
Felix Team

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HOLLAND RESEARCH SCHOOL  
OF MOLECULAR CHEMISTRY



## International Symposium on Small Particles and Inorganic Clusters

September 3<sup>rd</sup>-8<sup>th</sup>, 2023

Berlin, Germany

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!!!!!! NEW !!!!!

**Graduate Student seminar**

**September 2<sup>nd</sup>-3<sup>rd</sup> 2023**

All graduate student applicants  
can give an oral presentation.  
The best talks will be selected for  
Hot-Topic contributions at the ISSPIC.

abstract submission deadline: 30 June 2023!