CALCULATED VIBRATIONAL STATES OF OZONE UP TO DISSOCIATION

RICHARD DAWES, STEVE ALEXANDRE NDENGUE, Department of Chemistry, Missouri University of Science and Technology, Rolla, MO, USA; XIAO-GANG WANG, TUCKER CARRINGTON, Department of Chemistry, Queen's University, Kingston, ON, Canada; HUA GUO, Department of Chemistry and Chemical Biology, University of New Mexico, Albuquerque, NM, USA.

A new accurate global potential energy surface for the ground electronic state of ozone [J. Chem. Phys. 139, 201103 (2013)] was published fairly recently. The topography near dissociation differs significantly from previous surfaces, without spurious submerged reefs and corresponding van der Waals wells. This has enabled significantly improved descriptions of scattering processes, capturing the negative temperature dependence and large kinetic isotope effects in exchange reaction rates. The exchange reactivity was found to depend on the character of near-threshold resonances and their overlap with reactant and product wavefunctions, which in turn are sensitive to the potential. Here we present global "three-well" calculations of all bound vibrational states of three isotopic combinations of ozone for J = 0 and J = 1 with a focus on the character and density of highly excited states. The calculations were done using a parallel symmetry-adapted Lanczos method with the RTR code, enabling the use of as many as 64.8 million basis functions. Tunneling splittings and the pseudorotation isomerization path will be discussed.

