

## A ZERO-ORDER PICTURE OF THE INFRARED SPECTRUM FOR THE METHOXY RADICAL: ASSIGNMENT OF STATES

BRITTA JOHNSON, EDWIN SIBERT, *Department of Chemistry, University of Wisconsin–Madison, Madison, WI, USA.*

The ground  $\tilde{X}^2E$  vibrations of the methoxy radical have intrigued both experimentalists and theorists alike due to the presence of a conical intersection at the  $C_{3v}$  molecular geometry. This conical intersection causes methoxy's vibrational spectrum to be strongly influenced by Jahn-Teller vibronic coupling which leads to large amplitude vibrations and extensive mixing of the two lowest electronic states. This coupling combined with spin-orbit and Fermi couplings greatly complicates the assignments of states. Using the potential force field and calculated spectra of Nagesh and Sibert<sup>1,2</sup>, we assign quantum numbers to the infrared spectrum. When the zero-order states are the diabatic normal mode states, there is sufficient mode mixing that the normal mode quantum numbers are poor labels for the final states. We define a series of zero-order Hamiltonians which include additional coupling elements beyond the normal mode picture but still allow for the assignment of Jahn-Teller quantum numbers. In methoxy, the two lowest frequency *e* modes, the bend ( $q_5$ ) and the rock ( $q_6$ ), are the modes with the strongest Jahn-Teller coupling. In general, a zero-order Hamiltonian which includes first-order Jahn-Teller coupling in  $q_6$  is sufficient for most states of interest. Working in a representation which includes first-order Jahn-Teller coupling in  $q_6$ , we identify states in which additional coupling elements must be included; these couplings include first-order Jahn-Teller coupling in  $q_5$ , higher order Jahn-Teller coupling in  $q_5$  and  $q_6$ , and, in the dueterated case, Jahn-Teller coupling which is modulated by the corresponding *a* modes.

<sup>1</sup> Nagesh, J.; Sibert, E. L. *J. Phys. Chem. A* **2012**, *116*, 3846–3855.

<sup>2</sup> Lee, Y.F.; Chou, W.T.; Johnson, B.A.; Tabor, D.P. ; Sibert, E.L.; Lee, Y.P. *J. Mol. Spectrosc.* **2015**, *310*, 57-67.

<sup>2</sup> Barckholtz, T. A.; Miller, T. A. *Int. Revs. in Phys. Chem.* **1998**, *17*, 435–524.