

# PROBING THE METHYL TORSIONAL BARRIERS OF THE *E* AND *Z* ISOMERS OF BUTADIENYL ACETATE BY MICROWAVE SPECTROSCOPY

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The Fourier transform microwave spectra of the *E* and the *Z* isomer of butadienyl acetate have been measured in the frequency range from 2 to 26.5 GHz under molecular beam conditions. The most stable conformer of each isomer, in which all heavy atoms are located in a symmetry plane, was identified after analyzing the spectrum by comparison with results from quantum chemical calculations. The barrier to internal rotation of the acetyl methyl group was found to be  $149.1822(20) \text{ cm}^{-1}$  and  $150.2128(48) \text{ cm}^{-1}$  for the *E* and the *Z* isomer, respectively, which are similar to that of vinyl acetate <sup>a,b</sup>. A comparison between two theoretical approaches treating internal rotations, the rho axis method (using the program *BELGI-C<sub>s</sub>*) and combined axis method (using the program *XIAM*), is also performed.

Since several years we study the barriers to internal rotation of the acetyl methyl group in acetates,  $\text{CH}_3\text{--COOR}$ . Currently, we assume that all acetates can be divided into three classes. Class I contains  $\alpha,\beta$  saturated acetates, where the torsional barrier is always close to  $100 \text{ cm}^{-1}$ . Examples are a series of alkyl acetates such as methyl acetate and ethyl acetate. Class II contains  $\alpha,\beta$ -unsaturated acetates where the  $\text{C}=\text{C}$  double bond is located in the  $\text{COO}$  plane. This is the case of vinyl acetate and butadienyl acetate. Finally, in class III with isopropenyl acetate and phenyl acetate as two representatives,  $\alpha,\beta$ -unsaturated acetates, in which the double bond is not located in the  $\text{COO}$  plane, are collected. There, we observed a barrier height around  $135 \text{ cm}^{-1}$ . This observation will be discussed in details.

<sup>a</sup>B. Velino, A. Maris, S. Melandri, W. Caminati, J. Mol. Spectrosc. 2009, 256, 228

<sup>b</sup>H. V. L. Nguyen, A. Jabri, V. Van, and W. Stahl, J. Phys. Chem. A, 2014, 118, 12130